Greymouth Petroleum Turangi Ltd

Turangi-B Hydraulic Fracturing

Monitoring Programme

2019-2022

Technical Report 22-97

Working with people | caring for Taranaki

Taranaki Regional Council

Taranaki Regional Council
Private Bag 713
Stratford

ISSN: 1178-1467 (Online)
Document: 2997913 (Word)
Document: 3133741 (Pdf)
March 2023

Greymouth Petroleum Turangi Ltd

Turangi-B Hydraulic Fracturing

Monitoring Programme

2019-2022

Technical Report 22-97

Greymouth Petroleum Turangi Ltd

Turangi-B Hydraulic Fracturing

Monitoring Programme

2019-2022

Technical Report 22-97

Taranaki Regional Council
Private Bag 713
Stratford

ISSN: 1178-1467 (Online)
Document: 2997913 (Word)
Document: 3133741 (Pdf)
March 2023

Executive summary

Greymouth Petroleum Turangi Ltd (the Company) operates the Turangi-B hydrocarbon exploration site (wellsite) located on Turangi Road, Tikorangi located in the Parahaki catchment. This report outlines and discusses the results of the monitoring programme implemented by the Council in relation to hydraulic fracturing activities conducted by the Company at the wellsite over the period 14 November 2019 to 23 July 2021. The report also details the results of the monitoring undertaken and assesses the environmental effects of the Company's activities.

The programme of hydraulic fracturing undertaken by the Company at the Turangi-B wellsite included the hydraulic fracturing of four wells. The wells targeted for stimulation were the Turangi-8, Turangi-9, Turangi-10 and Turangi-11 wells.

During the monitoring period, the Company demonstrated an overall high level of environmental and a high level of administrative performance.

The programme of monitoring implemented by the Council in relation to these hydraulic fracturing activities spanned the 2018-2019, 2019-2020, 2020-2021 and 2021-2022 monitoring years. Monitoring included pre and post-discharge groundwater sampling in relation to discharges at the Turangi-B wellsite. In order to characterise the discharges and to determine compliance with consent conditions samples of hydraulic fracturing fluids, and fluids returning to the wellhead post-fracturing, were also collected by the Company and results of the physicochemical analysis undertaken were provided to the Council. This is the second monitoring report produced by the Council in relation to the hydraulic fracturing activities at the Turangi-B wellsite.

The monitoring carried out by the Council indicates that the hydraulic fracturing activities undertaken by the Company had no significant adverse effects on local groundwater or surface water resources. There were no unauthorised incidents recording non-compliance in respect of the resource consent held by the Company in relation to these activities or provisions in regional plans, during the period under review.

For reference, in the 2021-2022 year, consent holders were found to achieve a high level of environmental performance and compliance for 88% of the consents monitored through the Taranaki tailored monitoring programmes, while for another 10% of the consents, a good level of environmental performance and compliance was achieved

This report includes recommendations for the future monitoring of any hydraulic fracturing activities at the Turangi-B wellsite.

Table of contents

List of tables

List of figures

1 Introduction

1.1 Compliance monitoring programme reports and the Resource Management Act 1991

1.1.1 Introduction

This report outlines and discusses the results of the monitoring programme implemented by the Taranaki Regional Council (the Council) in relation to the programme of hydraulic fracturing undertaken by Greymouth Petroleum Turangi Ltd (the Company) at the Turangi-B wellsite, over the period 14 November 2019 to 23 July 2021. The report also assesses the Company's level of environmental performance and compliance with the resource consent held in relation to the activity.

The programme of hydraulic fracturing undertaken by the Company at the Turangi-B wellsite included the hydraulic fracturing of four wells. The wells targeted for stimulation were the Turangi-8, Turangi-9, Turangi-10 and Turangi-11 wells.

The programme of monitoring implemented by the Council in relation to these hydraulic fracturing activities spanned the 2018-2019, 2019-2020, 2020-2021 and 2021-2022 monitoring years. Monitoring included a mixture of groundwater, surface water and discharge monitoring components. This is the second monitoring report produced by the Council in relation to hydraulic fracturing activities at the Turangi-B wellsite.

1.1.2 Structure of this report

Section 1 of this report is a background section. It sets out general information about:

- the nature of the monitoring programme in place for the period under review; and
- a description of the activities and operations conducted at the Turangi-B wellsite.

Section 2 presents the results of monitoring during the period under review, including scientific and technical data.

Section 3 discusses the results, their interpretations, and their significance for the environment.

Section 4 presents recommendations to be implemented for the future monitoring of any hydraulic fracturing activities at the Turangi-B wellsite.

A glossary of common abbreviations and scientific terms, and a bibliography, are presented at the end of the report.

1.1.3 The Resource Management Act 1991 and monitoring

The Resource Management Act 1991 (RMA) primarily addresses environmental 'effects' which are defined as positive or adverse, temporary or permanent, past, present or future, or cumulative. Effects may arise in relation to:

a. the neighbourhood or the wider community around an activity, and may include cultural and social-economic effects;

b. physical effects on the locality, including landscape, amenity and visual effects;

c. ecosystems, including effects on plants, animals, or habitats, whether aquatic or terrestrial;

d. natural and physical resources having special significance (for example recreational, cultural, or aesthetic); and

e. risks to the neighbourhood or environment.

In drafting and reviewing conditions on discharge permits, and in implementing monitoring programmes, the Council is recognising the comprehensive meaning of 'effects' inasmuch as is appropriate for each activity. Monitoring programmes are not only based on existing permit conditions, but also on the obligations of the RMA to assess the effects of the exercise of consents. In accordance with Section 35 of the RMA, the Council undertakes compliance monitoring for consents and rules in regional plans, and maintains an overview of the performance of resource users and consent holders. Compliance monitoring, including both activity and impact monitoring, enables the Council to continually re-evaluate its approach and that of consent holders to resource management and, ultimately, through the refinement of methods and considered responsible resource utilisation, to move closer to achieving sustainable development of the region's resources.

1.1.4 Evaluation of environmental and administrative performance

Besides discussing the various details of the performance and extent of compliance by the Company, this report also assigns a rating as to each Company's environmental and administrative performance during the period under review. The rating categories are high, good, improvement required and poor for both environmental and administrative performance. The interpretations for these ratings are found in Appendix II.

For reference, in the 2021-2022 year, consent holders were found to achieve a high level of environmental performance and compliance for 88% of the consents monitored through the Taranaki tailored monitoring programmes, while for another 10% of the consents, a good level of environmental performance and compliance was achieved.[1]

1.2 Process description

1.2.1 Hydraulic fracturing

Hydraulic fracturing is a reservoir stimulation technique used to increase the flow of hydrocarbons to the surface. The primary objective of hydraulic fracturing is to increase the permeability of the target reservoir by creating numerous small, interconnected fractures, thus increasing the flow of hydrocarbons from the formation to a given well. The process of hydraulic fracturing has enabled companies to produce hydrocarbons at economically viable rates from extremely low permeability reservoirs and those that have become depleted using conventional production techniques.

The process of hydraulic fracturing involves the pumping of fluids and a proppant (medium-grained sand or small ceramic pellets) down a well, through a perforated section of the well casing, and into the target reservoir. The fluid mixture is pumped at a pressure that exceeds the fracture strength of the reservoir rock in order to create fractures. Once fractures have been initiated, pumping continues in order to force the fluid and proppant into the fractures created. The proppant is designed to keep the fractures open when the pumping is stopped. The placement of proppant into the fractures can be assisted by the use of cross-linked gels (gel fracturing), turbulent flow (slick-water fracturing), or the use of nitrogen gas.

[1] The Council has used these compliance grading criteria for more than 18 years. They align closely with the 4 compliance grades in the MfE Best Practice Guidelines for Compliance, Monitoring and Enforcement, 2018

1.2.1.1 Gel fracturing

Gel fracturing utilises cross-linked gel solutions, which are liquid at the surface but, when mixed, form long-chain polymer bonds and thus become viscous gels. These gels are used to transport the proppant into the formation. Once in the formation they 'break' back with time, temperature and the aid of gel breaking chemicals into a liquid state and are flowed back to surface, without disturbing the proppant which remains in place and enhances the flow of hydrocarbons back to the surface.

1.2.1.2 Slick water fracturing

Slick water fracturing utilises water based fracturing fluids with friction-reducing additives. The addition of the friction reducers allows the fracturing fluids and proppant to be pumped to the target zone at higher rates and reduced pressures, than when using water alone. The higher rate creates turbulence within the fluid column holding the proppant and enabling its placement into the open fractures and enhancing the flow of hydrocarbons back to the surface.

1.2.1.3 Nitrogen gas fracturing

Nitrogen gas assisted fracturing involves replacing some of the fluid used in the fracturing process with nitrogen gas, which can fracture rock at high pressures much like water. While nitrogen (N_2) is a gas at room temperature, it can be maintained in a liquid state through cooling and pressurisation. Nitrogen assisted fracturing can be beneficial from a production standpoint as inevitably during the fracturing process some of the water pumped down the well remains underground in the rock formation, which can block some of the small pores, inhibiting hydrocarbon recovery. The use of nitrogen gas reduces the amount of water required for each fracturing event. This also reduces the total concentration of chemical additives required and the volume of water returning to the surface that requires subsequent disposal.

1.2.2 The Turangi-B wellsite and hydraulic fracturing activities

The Turangi-B wellsite is located on Turangi Road Upper near Waitara and lies within the Parahaki catchment. The area surrounding the site is rural in nature and farming and forestry activities co-exist with active petroleum exploration and production operations. The location of the wellsite is illustrated in Figure 1. A summary of the hydraulic fracturing activities carried out by the Company at the Turangi-B wellsite during the period being reported is provided below in Table 1.

Table 1 Summary of hydraulic fracturing details

Well	Bore id.	Date range	Mid-point injection intervals (m TVDss)	Formation
Turangi-8	GND3107	14-11-2019 - 10-02-2020	3,416 – 4,397	Mangahewa/Kaimiro
Turangi-9	GND3108	10-03-2020 - 15-06-2020	3,408 – 4,426	
Turangi-10	GND3109	21-01-2021 - 04-03-2021	3,419 – 4,426	
Turangi-11	GND3110	13-03-2021 - 23-07-2021	3,404 – 4,425	

Figure 1 Location map

1.3 Resource consents

The Company holds one resource consent the details of which are summarised in Table 2 below. Summaries of the conditions attached to the permit are set out in Section 3 of this report.

A summary of the various consent types issued by the Council is included Appendix I, as is a copy of the permit held by the Company during the period under review.

Table 2 Resource consent held by the Company during the period under review

Consent number	Purpose of consent	Granted	Next review	Expires
7952-2	To discharge water based hydraulic fracturing fluids into land at depths greater than 3,410 m TVDss beneath the Turangi-B wellsite	9 June 2016	June 2023	01 June 2033

1.4 Monitoring programme

1.4.1 Introduction

Section 35 of the RMA sets obligations upon the Council to gather information, monitor and conduct research on the exercise of resource consents within the Taranaki region. The Council is also required to assess the effects arising from the exercising of these consents and report upon them.

The Council may therefore make and record measurements of physical and chemical parameters, take samples for analysis, carry out surveys and inspections, conduct investigations and seek information from consent holders.

The monitoring programme for the Turangi-B wellsite consisted of four primary components.

1.4.2 Programme liaison and management

There is generally a significant investment of time and resources by the Council in:

- ongoing liaison with resource consent holders over consent conditions and their interpretation and application;
- in discussion over monitoring requirements;
- preparation for any consent reviews, renewals or new consent applications;
- advice on the Council's environmental management strategies and content of regional plans; and
- consultation on associated matters.

1.4.3 Assessment of data submitted by the consent holder

As required by the conditions of consent 7952-2, the Company submitted pre and post fracturing discharge reports to the Council for each well fractured during the period under review. Pre-fracturing discharge reports provide an outline of the proposed fracturing operations in relation to each well, while post fracturing reports confirm details of what actually occurred. The specific range of information required in each report is stipulated in the conditions of the consent.

1.4.4 Physicochemical sampling

1.4.4.1 Groundwater

As a general principle, all existing bores or wells within a 1 km radius of a hydraulic fracturing activity are assessed for their suitability for sampling (or otherwise) and included in the monitoring programme for the wellsite.

The survey of existing sites resulted in a total of 15 potential groundwater sampling locations being identified. Assessment of the sites indicated that they were all either greater than 500 m from the wellsite or were very shallow monitoring bores, previously installed by the company for monitoring various activities onsite.

As there were no suitable monitoring sites identified, consent conditions required the Company to install a site specific monitoring bore. The new bore installed by the Company is the sole groundwater monitoring site included in the monitoring programme. A summary of bore details are included in Table 3 below.

Samples of groundwater were obtained pre-fracturing to provide a baseline reference of groundwater composition and a further six rounds of sampling were carried out during and following completion of the activities.

Groundwater samples are collected following standard groundwater sampling methodologies and generally in accordance with the National Environmental Monitoring Standards (NEMS) for discrete groundwater quality sampling (2019). All samples were transported to Hill Laboratories Ltd (Hills) for analysis following standard chain of custody procedures.

Table 3 Details of groundwater sites included in the monitoring programme

Monitoring site	Easting (NZTM)	Northing (NZTM)	Distance from wellsite (m)	Total depth (m)	Screened/open interval (m)	Aquifer
GND3061	1713538	5682554	<50	33	23-32	Marine terraces north

1.4.4.2 Hydraulic fracturing and return fluids

In addition to the sampling of local groundwater, representative samples of the hydraulic fracturing fluid and reservoir fluids produced back to the wellhead immediately following each fracturing event (return fluids) are obtained for analysis.

Samples of return fluids were collected at regular intervals during the flow-back period. Return fluids are comprised of a mixture of hydraulic fracturing fluids and formation fluids produced from the target reservoir, following the completion of the hydraulic fracturing process. The relative concentrations of each contributing fluid type change as the volume of fluid produced from the well increases. Immediately following the opening of the well post fracturing, a high proportion of the fluid returning to the wellhead is fluid injected during the hydraulic fracturing process. As the volume of fluid produced from the well increases, the proportion of hydraulic fracturing fluid reduces in relation to formation fluids. The individual samples of return fluid are generally combined in a composite sample for laboratory analysis. Composites are designed to provide a representative sample of fluids returning to the wellhead over the entire flow-back period.

All samples are collected and sent by the Company to Hills for analysis. Results are provided to the Council on request or in the post fracturing report provided following each fracturing event.

1.4.5 Surface water quality monitoring

The Parahaki Stream is located to the south west of the Turangi-B wellsite (Figure 1). Following a survey of the area no suitable surface water monitoring sites were identified down-gradient of the well site discharges and the estimated location of groundwater/subsurface drainage from the discharge area. Therefore no surface water monitoring was required for inclusion in the Hydraulic Fracturing monitoring programme.

Following a temporary change in wellsite discharge from land to water at the wellsite, three surface water sampling sites were established in 2021 and two bio-monitoring surveys undertaken. The first on 11 June 2021 and the second on 19 January 2022. The results of these surveys will be reported on as part of the wider wellsite monitoring programme and have not been included in this report.

Table 4 Details of surface water sites included in the monitoring programme

Site code	Eastings	Northings	Distance from wellsite (m)
PRH000046	1713407	5682306	10m upstream of Turangi-B wellsite discharge
PRH000047	1713402	5682342	25m downstream of Turangi-B wellsite discharge
PRH000048	1713383	5682392	120m downstream of Turangi-B wellsite discharge

2 Results

2.1 Consent holder submitted data

The conclusions from the Turangi-B post fracturing discharge reports are summarised as follows:

2.1.1 Turangi-8 post fracturing discharge report

- A total of seven zones were fractured over the period 14 November 2019 to 10 February 2020 at mid-point depths between 3,416 to 4,396 m TVDss.
- A total of 11,240 bbls (1,787 m^3) of liquid was discharged across the seven fractured units. The total proppant weight was 335 tonnes (739,619 lbs).
- The majority of fluid injected was returned from the well over the flow-back period with any remaining fluid expected to be returned during subsequent operations.
- A total of 325 tonnes (717,603 lbs) of proppant was estimated to have remained within the formation following flow-back.
- One screen out occurred during hydraulic fracturing of the Turangi-8 well. The screen-out occurred during Treatment 7 and resulted in under-placement of some proppant. There were no implications in compliance with conditions 1 and 5 of the consent.
- All return fluid from the Turangi-8 fracturing operations were disposed of at the Company's Kaimiro-G wellsite under the Company's deep well injection consent.
- Pressure testing was undertaken of all surface equipment, including flow lines and the wellhead, prior to injection.
- There was no escape of fluids during hydraulic fracturing operations.

2.1.2 Turangi-9 post fracturing discharge report

- A total of eight zones were fractured over the period 10 March 2020 to 15 June 2020 at mid-point depths between 3,408 to 4,426 m TVDss.
- A total of 14,326 bbls (2,278 m^3) of liquid was discharged across the seven fractured units. The total proppant weight was 390 tonnes (859,428 lbs).
- The majority of fluid injected was returned from the well over the flow-back period with any remaining fluid expected to be returned during subsequent operations.
- A total of 385 tonnes (849,484 lbs) of proppant was estimated to have remained within the formation following flow-back.
- One screen out occurred during hydraulic fracturing of the Turangi-9 well. The screen-out occurred during Treatment 5 and resulted in under-placement of some proppant. There were no implications in compliance with conditions 1 and 5 of the consent.
- All return fluid from the Turangi-9 fracturing operations were disposed of at the Company's Kaimiro-G wellsite under the Company's deep well injection consent.
- Pressure testing was undertaken of all surface equipment, including flow lines and the wellhead, prior to injection.
- There was no escape of fluids during hydraulic fracturing operations.

2.1.3 Turangi-10 post fracturing discharge report

- A total of six zones were fractured over the period 21 January 2021 to 04 March 2021 at mid-point depths between 3,419 to 4,426 m TVDss.

- A total of 10,825 bbls (1,721 m^3) of liquid was discharged across the seven fractured units. The total proppant weight was 304 tonnes (671,272 lbs).
- The majority of fluid injected was returned from the well over the flow-back period with any remaining fluid expected to be returned during subsequent operations.
- A total of 281 tonnes (618,447 lbs) of proppant was estimated to have remained within the formation following flow-back.
- Treatment 2, Treatment 3 and Treatment 4 screened out during hydraulic fracturing of the Turangi-10 well resulting in under-placement of some proppant. There were no implications in compliance with conditions 1 and 5 of the consent.
- All return fluid from the Turangi-10 fracturing operations were disposed of at the Company's Kaimiro-G wellsite under the Company's deep well injection consent.
- Pressure testing was undertaken of all surface equipment, including flow lines and the wellhead, prior to injection.
- There was no escape of fluids during hydraulic fracturing operations.

2.1.4 Turangi-11 post fracturing discharge report

- A total of seven zones were fractured over the period 13 March 2021 to 23 July 2021 at mid-point depths between 3,404 to 4,425 m TVDss.
- A total of 11,988 bbls (1,908 m^3) of liquid was discharged across the seven fractured units. The total proppant weight was 336 tonnes (740,277 lbs).
- The majority of fluid injected was returned from the well over the flow-back period with any remaining fluid expected to be returned during subsequent operations.
- A total of 329 tonnes (725,680 lbs) of proppant was estimated to have remained within the formation following flow-back.
- Treatment 1 screened out during hydraulic fracturing of the Turangi-11 well resulting in under-placement of some proppant. There were no implications in compliance with conditions 1 and 5 of the consent.
- All return fluid from the Turangi-10 fracturing operations were disposed of at the Company's Kaimiro-G wellsite under the Company's deep well injection consent.
- Pressure testing was undertaken of all surface equipment, including flow lines and the wellhead, prior to injection.
- There was no escape of fluids during hydraulic fracturing operations.

2.2 Physicochemical sampling

2.2.1 Groundwater

Hydraulic fracturing activities commenced at the Turangi-B wellsite on 14 November 2019 and continued until 23 July 2021. A pre-fracturing baseline sample was collected on 8 November 2019 following installation of a site specific monitoring bore. Post fracturing samples were collected at various intervals following commencement of the activities which including hydraulic fracturing in four wells and spanned 22 months. Interim samples were collected on 24 February 2020, 29 July 2020, 10 November 2020, 25 February 2021, 9 July 2021 and 23 December 2021. A post fracturing sample was collected on 14 July 2022.

Methane concentrations >1 g m^3 were reported in all samples both pre and post-hydraulic fracturing activities and can occur as a result of biogenic processes in sulphate depleted groundwater systems. To determine whether the source of the methane was biogenic or thermogenic, the majority of samples were

sent to Geological and Nuclear Sciences (GNS) for carbon 13 isotope analysis. One sample could not be analysed as the bottles were damaged during transport. The presence of carbon 13 isotopes concentrations at less than -50‰ indicate a thermogenic deep gas source and concentrations greater than -50‰ a shallow biogenic gas source. Carbon 13 concentrations ranged between -40.1 to -72.6 ‰ indicating a mixed biogenic/thermogenic source of methane at the site.

Toluene was also reported in three of the eight samples at trace concentrations. The presence of trace toluene and methane is not indicative of any significant change in groundwater quality and is likely a result of groundwater interaction with peat deposits, which were identified at the site during installation of the bore.

Overall, samples demonstrate relatively narrow ranges in analyte concentrations over time. The subtle variation in analyte concentrations are a result of natural seasonal fluctuation and sampling variability. The results of the laboratory analysis indicate there have been no significant changes in groundwater composition over the period monitored.

A summary of the results for groundwater samples taken in relation to the hydraulic fracturing activities compared to baseline is included in Table 4. The certificates of analysis for the review period are included in Appendix III.

Table 5 Results of groundwater sampling carried out in relation to the Turangi-B fracturing events

Sample date	unit	8/11/2019	24/02/2020	29/07/2020	10/11/2020	25/02/2021	9/07/2021	23/12/2021	14/06/2022
Sample time	-	12:15	13:20	11:00	10:25	11:05	11:20	9:45	12:45
Sample id. TRC	-	TRC193990	TRC200746	TRC202067	TRC203765	TRC210881	TRC212297	TRC214610	TRC226761
pH	pH	7.5	7.4	6.9	7.1	7.4	6.9	7.1	7.1
Temperature	°C	18.4	18.6	15.5	17.1	17.1	15.0	17.0	15.0
Total alkalinity	g/m^3 $CaCO_3$	182	180	155	166	167	159	149	141
Bicarbonate	g/m^3 HCO_3	220	220	189	200	200	194	182	172
Total hardness	g/m^3 $CaCO_3$	146	117	122	117	108	125	112	105
Electrical conductivity	µS/cm	408	379	342	357	360	361	327	312
Total dissolved solids	g/m^3	240	250	250	260	220	240	240	260
Dissolved calcium	g/m^3	36	27	30	28	24	30	26	24
Chloride	g/m^3	18.3	16.4	16.2	15.5	16.3	16.4	17.1	16.6
Dissolved magnesium	g/m^3	13.4	11.9	11.5	11.6	12.0	12.3	11.4	10.8
Dissolved potassium	g/m^3	9.7	8.0	6.0	6.4	7.0	5.2	5.5	4.2
Dissolved sodium	g/m^3	30	31	21	23	28	21	21	19
Nitrite	g/m^3 N	< 0.002	0.003	0.005	< 0.02	0.004	< 0.02	< 0.02	< 0.02
Nitrate	g/m^3 N	0.003	0.03	< 0.002	< 0.02	0.012	< 0.02	< 0.02	< 0.02
Nitrate & nitrite	g/m^3 N	0.003	0.034	0.003	< 0.02	0.016	< 0.02	< 0.02	< 0.02
Sulphate	g/m^3	4.9	< 0.5	0.5	< 0.5	< 0.5	1.0	< 0.5	< 0.5
Dissolved barium	g/m^3	0.034	0.029	0.044	0.044	0.033	0.064	0.047	0.051
Bromide	g/m^3	0.11	0.09	< 0.05	0.09	0.06	0.09	0.09	0.07
Dissolved copper	g/m^3	< 0.0005	< 0.0005	< 0.0005	< 0.0005	< 0.0005	< 0.0005	< 0.0005	< 0.0005
Dissolved iron	g/m^3	0.07	3.7	8.0	9.5	4.5	10.8	8.3	9.4
Dissolved manganese	g/m^3	0.25	0.28	0.45	0.48	0.23	0.48	0.37	0.38
Dissolved mercury	g/m^3	< 0.00008	< 0.00008	< 0.00008	< 0.00008	< 0.00008	< 0.00008	< 0.00008	< 0.00008
Dissolved Nickel	g/m^3	0.0011	0.0006	0.0006	< 0.0005	< 0.0005	0.0005	< 0.0005	0.0006
Dissolved zinc	g/m^3	< 0.0010	0.0027	0.013	0.0062	0.0052	0.0044	0.0113	0.0116

Sample date	unit	8/11/2019	24/02/2020	29/07/2020	10/11/2020	25/02/2021	9/07/2021	23/12/2021	14/06/2022
Sample time	-	12:15	13:20	11:00	10:25	11:05	11:20	9:45	12:45
Sample id. TRC	-	TRC193990	TRC200746	TRC202067	TRC203765	TRC210881	TRC212297	TRC214610	TRC226761
Ethylene glycol	g/m^3	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4
Propylene glycol	g/m^3	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4
Methanol	g/m^3	< 2	< 2	< 2	< 2	< 4	< 2	< 2	< 2
Benzene	g/m^3	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010
Toluene	g/m^3	0.0023	0.0014	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	0.0014
Ethylbenzene	g/m^3	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010
m-Xylene	g/m^3	< 0.002	< 0.002	< 0.002	< 0.002	< 0.002	< 0.002	< 0.002	< 0.002
o-Xylene	g/m^3	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010	< 0.0010
Formaldehyde	g/m^3	< 0.02	< 0.02	0.02	< 0.02	< 0.02	0.02	< 0.02	< 0.02
Ethane	g/m^3	< 0.003	< 0.003	< 0.003	< 0.003	< 0.003	< 0.003	< 0.003	< 0.003
Ethylene	g/m^3	< 0.004	< 0.004	< 0.003	< 0.004	< 0.004	< 0.004	< 0.004	< 0.004
Methane	g/m^3	7.5	8.2	7.7	11.5	8.3	14.2	11.2	11.7
C7-C9 hydrocarbons	g/m^3	< 0.10	< 0.10	< 0.10	< 0.10	< 0.10	< 0.10	< 0.10	< 0.10
C10-C14 hydrocarbons	g/m^3	< 0.2	< 0.2	< 0.2	< 0.2	< 0.2	< 0.2	< 0.2	< 0.2
C15-C36 hydrocarbons	g/m^3	< 0.4	< 0.4	< 0.4	< 0.4	< 0.4	< 0.4	< 0.4	< 0.4
Total hydrocarbons	g/m^3	< 0.7	< 0.7	< 0.7	< 0.7	< 0.7	< 0.7	< 0.7	< 0.7
δ13C value	‰	-55.7	-67.3	-60.4	-72.6	Not analysed	-46.6	-40.1	-70.8

2.2.2 Hydraulic fracturing and return fluids

The results of the analyses carried out on samples of the hydraulic fracturing fluid used in the stimulation of the Turangi-8, Turangi-9, Turangi-10 and Turangi-11 wells are shown below in Table 5. The certificates of analysis are included in Appendix IV.

The results of the analyses carried out on the return fluid samples obtained following the hydraulic fracturing of the Turangi-8, Turangi-9, Turangi-10 and Turangi-11 wells are summarised below in Table 6. The certificates of analysis are included in Appendix III. The results demonstrate the variability of groundwater composition and hydrocarbon concentrations during flow-back. The relatively high levels of chloride, sodium and hydrocarbons in each sample indicate that the composite samples prepared contained a greater proportion of reservoir fluids than hydraulic fracturing fluids introduced during the fracturing activities, which are comprised predominantly of freshwater.

Table 6 Results of hydraulic fracturing fluid sampling

Well	Unit	Turangi 8		Turangi 9		Turangi 10		Turangi 11	
Parameter	Range	Min	Max	Min	Max	Min	Max	Min	Max
Ethylene glycol *	g/m^3	<4	<20	<4	<20	<20	<400	<20	<20
Propylene Glycol *	g/m^3	<4	<20	<4	<20	<20	<20	<20	<20
Methanol*	g/m^3	<2	<2	<20	<20	<2	<40	<20	2
Benzene	g/m^3	<0.010	0.013	0.0016	0.024	<0.010	<0.010	0.0015	0.022
Toluene	g/m^3	<0.010	0.029	0.0023	0.05	<0.010	0.016	0.0027	0.03
Ethylbenzene	g/m^3	<0.010	<0.010	<0.0010	<0.011	<0.010	<0.010	<0.0010	<0.010
m&p-Xylene	g/m^3	<0.02	<0.02	<0.002	<0.07	<0.02	<0.02	<0.0020	0.0300
o-Xylene	g/m^3	<0.010	<0.012	<0.001	<0.021	<0.010	<0.010	<0.001	0.0012
C7 - C9	g/m^3	0.4	8.2	<0.4	1.3	<0.6	2	<0.4	1.7
C10 - C14	g/m^3	900	2,900	189	4,200	1,450	6,500	470	4,300
C15 - C36	g/m^3	166	1,120	61	470	89	830	64	183
Total hydrocarbons	g/m^3	1,190	3,200	250	4,700	1,590	7,400	530	4,400

*Note * Depending on the viscosity of the sample received at the laboratory, samples may require dilution prior to analysis which results in higher detection limits.*

Table 7 Results of hydraulic fracturing return fluid sampling

Well	Unit	Turangi-8		Turangi-9		Turangi-10		Turangi-11	
Parameter	Range	Min	Max	Min	Max	Min	Max	Min	Max
pH	pH	5.7	8	6.9	7.5	6.5	7.7	6.4	9.7
Total Alkalinity	g/m^3 as $CaCO_3$	280	2,800	560	3,100	660	5,000	106	4,400
Total Hardness	g/m^3 as $CaCO_3$	88	400	51	560	70	410	45	470
Electrical Conductivity	mS/m	474	1,824	552	2,260	642	1,850	104	6,040
Total Dissolved Solids	g/m^3	4,300	12,100	4,900	15,200	5,500	13,600	1,620	21,000
Dissolved Barium	g/m^3	1.29	38	1.99	65	1.9	56	0.182	72
Dissolved Bromine	g/m^3	2.3	16.5	1.9	19	4.3	16.8	< 0.5	27
Dissolved Calcium	g/m^3	25	126	14	182	21	130	12	156
Dissolved Copper	g/m^3	<0.005	0.076	< 0.005	0.065	<0.005	0.025	0.005	0.27
Dissolved Iron	g/m^3	1.18	17.1	2.1	6.7	0.15	15.8	0.35	10.6
Dissolved Magnesium	g/m^3	1	20	4	25	4	21	3	21
Dissolved Manganese	g/m^3	2.4	8.5	0.88	4.4	0.93	3.3	0.016	5.2

Well	Unit	Turangi-8		Turangi-9		Turangi-10		Turangi-11	
Parameter	Range	Min	Max	Min	Max	Min	Max	Min	Max
Total Nickel	g/m^3	0.035	1.27	0.089	1.08	0.084	0.57	< 0.032	0.41
Total Potassium	g/m^3	50	290	96	270	108	340	9	330
Total Sodium	g/m^3	930	3,900	1,130	4,900	1,240	3,900	49	5,300
Total Sulphur	g/m^3	21	48	10	42	19	<60	10	< 60
Total Zinc	g/m^3	0.037	0.151	0.06	0.41	0.056	5.6	0.036	2.4
Chloride	g/m^3	790	6,200	1,030	6,500	1,420	5,200	175	8,000
Nitrite	g/m^3 N	<0.010	<0.010	<0.10	0.15	<0.010	<0.10	<0.010	0.34
Nitrate	g/m^3 N	<0.010	<0.010	<0.10	0.14	<0.010	<0.10	<0.010	0.05
Nitrate + Nitrite	g/m^3 N	<0.10	<0.10	<0.10	0.16	0.05	<0.10	< 0.10	0.35
Sulphate	g/m^3	64	145	31	125	56	<160	29	<160
Ethylene glycol	g/m^3	<4	43	<20	<20	45	490	<20	9300
Propylene glycol	g/m^3	<4	<20	<20	<20	<20	<20	<20	25
Methanol	g/m^3	<2	139	<20	410	<20	<110	<20	4900
Benzene	g/m^3	2.3	33	<0.010	19.8	6.5	51	7.6	35
Toluene	g/m^3	2.6	81	<0.010	21	7.9	240	6.2	103
Ethylbenzene	g/m^3	0.38	9.5	0.038	2.3	0.54	40	0.23	14.8
m&p-Xylene	g/m^3	2.2	61	2	10.1	3.1	370	1.21	103
o-Xylene	g/m^3	0.89	17.3	0.014	3.8	0.95	87	0.5	30
Formaldehyde	g/m^3	<0.15	<0.15	<0.15	<0.15	<0.15	3.7	<0.15	7.7
C7 - C9	g/m^3	21	690	20	620	68	5,400	16	1,080
C10 - C14	g/m^3	95	11,170	117	1,860	440	9,600	52	2,200
C15 - C36	g/m^3	93	1,200	110	1,470	260	13,900	52	1,150
TPH	g/m^3	280	3,100	390	3,900	810	29,000	166	4,500

*Note * Depending on the viscosity of the sample received at the laboratory, samples may require dilution prior to analysis which results in higher detection limit.*

2.3 Incidents, investigations and interventions

The monitoring programme for the year was based on what was considered to be an appropriate level of monitoring, review of data, and liaison with the Company. During the year matters may arise which require additional activity by the Council, for example provision of advice and information, or investigation of potential or actual causes of non-compliance or failure to maintain good practices. A pro-active approach that in the first instance avoids issues occurring is favoured.

The Council operates and maintains a register of all complaints or reported and discovered excursions from acceptable limits and practices, including non-compliance with consents, which may damage the environment. The incident register includes events where the consent holder concerned has itself notified the Council. The register contains details of any investigation and corrective action taken.

Complaints may be alleged to be associated with a particular site. If there is potentially an issue of legal liability, the Council must be able to prove by investigation that the identified company is indeed the source of the incident (or that the allegation cannot be proven).

During the period under review, the Council was not required to undertake significant additional investigations and interventions, or record incidents, in association with the Company's conditions in resource consents or provisions in Regional Plans.

3 Discussion

3.1 Environmental effects of exercise of consents

Four wells (Turangi-8, Turangi-9, Turangi-10 andTurangi-11) were stimulated by hydraulic fracturing at the Turangi-B wellsite during the period 14 November 2019 to 23 July 2021.

The monitoring programme carried out by the Council in relation to the fracturing events undertaken included pre and post fracturing sampling at one groundwater monitoring site in the vicinity of the Turangi-B wellsite. The results of post fracturing groundwater sampling carried out generally showed only very minor variations in water composition in comparison to baseline results. The minor variations in analytes are a result of natural variations in water composition.

There was no surface water monitoring undertaken in relation to hydraulic fracturing activities at the wellsite.

In summary, the monitoring carried out by the Council during the period being reported indicated that the hydraulic fracturing activities undertaken by the Company at the Turangi-B wellsite have had no significant adverse effects on local groundwater or surface water resources.

3.2 Evaluation of performance

A tabular summary of the consent holder's compliance record for the year under review is set out in Table 8.

Table 8 Summary of performance for consent 7952-2

Purpose: ***To discharge water based hydraulic fracturing fluids into land at depths greater than 3,200 metres true vertical depth subsea (TVDss) beneath the Turangi-B wellsite***		
Condition requirement	**Means of monitoring during period under review**	**Compliance achieved?**
1. Any discharge shall occur below 3,200 m TVDss	Assessment of consent holder submitted data	Yes
2. No discharge shall occur after 1 June 2028	Assessment of consent holder submitted data	N/A
3. Undertake micro seismic monitoring for events within 1 km of the Inglewood fault	Notification and post fracturing report	N/A
4. Monitoring and reporting of seismic events within 5 km of any discharge location	Notification and post fracturing report	Yes
5. Actions to be taken following the occurrence of any event described in condition 3 and 4	Notification under condition 3 and 4	N/A
6. Exercise of consent shall not result in any contaminants reaching any useable freshwater	Results of groundwater monitoring	Yes
7. Consent holder shall undertake sampling programme	Development and certification of a monitoring programme	Yes
8. If no suitable bores exist within 500 m of the wellsite, a monitoring bore may need to be installed	Inspection of bores	Yes

Purpose: *To discharge water based hydraulic fracturing fluids into land at depths greater than 3,200 metres true vertical depth subsea (TVDss) beneath the Turangi-B wellsite*		
Condition requirement	**Means of monitoring during period under review**	**Compliance achieved?**
9. Sampling programme shall follow recognised field procedures and be analysed for a specified range of chemical parameters	Development and certification of a monitoring programme and assessment of results	Yes
10. All sampling to be carried out in accordance with a certified Sampling and Analysis Plan	Development and certification of a Sampling and Analysis Plan	Yes
11. Well and equipment pressure testing to be carried out prior to any hydraulic fracturing programme commencing	Assessment of consent holder submitted data	Yes
12. A pre-fracturing discharge report is to be provided to the Council 14 days prior to discharge	Pre-fracturing discharge report received	Yes
13. Consent holder shall notify the Council of hydraulic fracturing discharge	Notification received	Yes
14. A post fracturing discharge report is to be provided to the Council within 90 days of any commencement	Post fracturing discharge report received	Yes
15. For programs including multiple hydraulic fracturing discharges, more than one 'Post-fracturing discharge report' may be required	Reports received via email	Yes
16. A review of the seismic monitoring data to be provided to the Council within 6 months of the commencement of any hydraulic fracturing	Seismic monitoring report received	Yes
17. The reports outlined in conditions 12, 14 and 15 must be emailed to consents@trc.govt.nz	Report received by email	Yes
18. The consent holder shall provide access to a location where samples of hydraulic fracturing fluids and return fluids can be obtained by the Council officers	Access provided	Yes
19. Consent holder to adopt best practicable option at all times	Site inspections, sampling and assessment of consent holder submitted data	Yes
20. No hydrocarbon based hydraulic fracturing fluid shall be discharged	Assessment of consent holder submitted data and sampling of fracturing fluid	Yes
21. Lapse clause	Receive notice of exercise of consent	Yes
22. Review condition	N/A	N/A

Purpose: *To discharge water based hydraulic fracturing fluids into land at depths greater than 3,200 metres true vertical depth subsea (TVDss) beneath the Turangi-B wellsite*		
Condition requirement	**Means of monitoring during period under review**	**Compliance achieved?**
Overall assessment of environmental performance and compliance in respect of this consent		**High**
Overall assessment of administrative performance and compliance in respect of this consent		**High**

N/A = not applicable

During the monitoring period, the Company demonstrated a high level of environmental and high level of administrative performance with the resource consent as defined in Appendix II.

3.3 Alterations to monitoring programmes for future hydraulic fracturing activities

In designing and implementing the monitoring programmes for air/water discharges in the region, the Council has taken into account:

- the extent of information already made available through monitoring or other means to date;
- its relevance under the RMA;
- the Council's obligations to monitor consented activities and their effects under the RMA;
- the record of administrative and environmental performances of the consent holder; and
- reporting to the regional community.

The Council also takes into account the scope of assessments required at the time of renewal of permits, and the need to maintain a sound understanding of industrial processes within Taranaki exercising resource consents.

It is proposed that the range of monitoring carried out in relation to the hydraulic fracturing activities undertaken by the Company be replicated for any future fracturing events at the Turangi-B wellsite.

Recommendations to this effect are included in Section 4 of this report.

It should be noted that the proposed programme represents a reasonable and risk-based level of monitoring for the site in question. The Council reserves the right to subsequently adjust the programme from that initially prepared, should the need arise if potential or actual non-compliance is determined at any time during future monitoring periods.

3.4 Exercise of optional review of consent

Resource consent 7952-2 provides for an optional review of the consent in June 2023. Condition 22 allows the Council to review the consent, for the purpose of:

a. ensuring that the conditions are adequate to deal with any significant adverse effects on the environment arising from the exercise of this consent, which were either not foreseen at the time the application was considered or which it was not appropriate to deal with at the time; and/or

b. further specifying the best practicable option as required by condition 19; and/or

c. ensuring hydraulic fracturing operations appropriately take into account any best practice guidance published by a recognised industry association or environmental regulator.

Based on the results of monitoring in the period under review, it is considered that there are no grounds that require a review to be pursued or grounds to exercise the review option.

4 Recommendations

1. THAT in the first instance, the range of monitoring carried out during the reporting period in relation to the Company's hydraulic fracturing activities be replicated for any future fracturing events at the Turangi-B wellsite.
2. THAT should there be issues with environmental or administrative performance in future periods, monitoring may be adjusted to reflect any additional investigation or intervention as found necessary.
3. THAT the option for a review of resource consents in June 2023, as set out in condition 22 of the consent not be exercised.

Glossary of common terms and abbreviations

The following abbreviations and terms may be used within this report:

Biomonitoring	Assessing the health of the environment using aquatic organisms.
bbls	Barrel. Unit of measure used in the oil and gas industry (equivalent to approximately 159 L).
Conductivity	An indication of the level of dissolved salts in a sample, usually measured at 25°C and expressed in µS/cm.
DO	Dissolved oxygen.
E.coli	Escherichia coli, an indicator of the possible presence of faecal material and pathological micro-organisms. Usually expressed as colony forming units per 100 millilitre sample.
EPT	Ephemeroptera (mayfly), Plecoptera (stonefly) and Trichoptera (caddisfly) which are macroinvertebrates sensitive to pollution.
Fresh	Elevated flow in a stream, such as after heavy rainfall.
g/m^3	Grams per cubic metre, and equivalent to milligrams per litre (mg/L). In water, this is also equivalent to parts per million (ppm), but the same does not apply to gaseous mixtures.
Incident	An event that is alleged or is found to have occurred that may have actual or potential environmental consequences or may involve non-compliance with a consent or rule in a regional plan. Registration of an incident by the Council does not automatically mean such an outcome had actually occurred.
Intervention	Action/s taken by Council to instruct or direct actions be taken to avoid or reduce the likelihood of an incident occurring.
Investigation	Action taken by Council to establish the circumstances/events surrounding an incident including any allegations of an incident.
L/s	Litres per second.
Macroinvertebrate	An invertebrate that is large enough to be seen without the use of a microscope.
masl	Metres above sea level.
MCI	Macroinvertebrate community index; a numerical indication of the state of biological life in a stream that takes into account the sensitivity of the taxa present to organic pollution in stony habitats.
mS/m	Millisiemens per metre.
m^3	Cubic metre (1,000 L).
NZTM	New Zealand Transverse Mercator coordinates.
pH	A numerical system for measuring acidity in solutions, with 7 as neutral. Numbers lower than 7 are increasingly acidic and higher than 7 are increasingly alkaline.
Physicochemical	Measurement of both physical properties (e.g. temperature, clarity, density) and chemical determinants (e.g. metals and nutrients) to characterise the state of an environment.
Resource consent	Refer Section 87 of the RMA. Resource consents include land use consents (refer Sections 9 and 13 of the RMA), coastal permits (Sections 12, 14 and 15), water permits (Section 14) and discharge permits (Section 15).
RMA	Resource Management Act 1991 and including all subsequent amendments.

Screen Out	A condition that occurs when the solids carried in a treatment fluid, such as proppant in a fracture fluid, create a bridge across the perforations or similar restricted flow area. This creates a sudden and significant restriction to fluid flow that causes a rapid rise in pump pressure.
SQMCI	Semi quantitative macroinvertebrate community index.
TVDss	True vertical depth sub-sea.
µS/cm	Microsiemens per centimetre.
Workover	The repair or stimulation of an existing production well for the purpose of restoring, prolonging or enhancing the production of hydrocarbons.

For further information on analytical methods, contact an Environment Quality Manager.

Bibliography and references

Baker Hughes at http://blogs.bakerhughes.com/reservoir/2012/02/20/hydraulic-fracturing-an-environmentally-responsible-technology-for-ensuring-our-energy-future-part-iii-of-iii

Geology.Com at http://geology.com/energy/hydraulic-fracturing-fluids/

Ministry for the Environment. 2018 Guidelines for compliance monitoring and enforcement under the Resource Management Act 1991. Wellington: Ministry for the Environment

Greymouth Petroleum Ltd (2015) Turangi-B Hydraulic Fracturing Monitoring Programme Report 2013-2015 Technical Report 2014-115. Frodo number #1487847

Greymouth Petroleum Turangi Ltd (2019) Turangi-8 Pre-Fracturing Report. Frodo id #2328731

Greymouth Petroleum Turangi Ltd (2019) Turangi-B water quality monitoring programme. Frodo id #2330433

Greymouth Petroleum Turangi Ltd (2020) Turangi-8 Post-fracturing Report. Frodo id #2486541

Greymouth Petroleum Turangi Ltd (2019) Turangi-9 Pre-Fracturing Report. Frodo id #2501298

Greymouth Petroleum Turangi Ltd (2020) Turangi-9 Post-fracturing Report. Frodo id #2572456

Greymouth Petroleum Turangi Ltd (2020) Turangi-10 Pre-Fracturing Report. Frodo id #2673243

Greymouth Petroleum Turangi Ltd (2021) Turangi-11 Pre-Fracturing Report. Frodo id #2700756

Greymouth Petroleum Turangi Ltd (2021) Turangi-10 Post-fracturing Report. Frodo id #2751577

Greymouth Petroleum Turangi Ltd (2021) Turangi-11 Post-fracturing Report. Frodo id #2857350

Appendix I

Resource consent held by Greymouth Petroleum Ltd

(For a copy of the signed resource consent
please contact the TRC Consents department)

Water abstraction permits

Section 14 of the RMA stipulates that no person may take, use, dam or divert any water, unless the activity is expressly allowed for by a resource consent or a rule in a regional plan, or it falls within some particular categories set out in Section 14. Permits authorising the abstraction of water are issued by the Council under Section 87(d) of the RMA.

Water discharge permits

Section 15(1)(a) of the RMA stipulates that no person may discharge any contaminant into water, unless the activity is expressly allowed for by a resource consent or a rule in a regional plan, or by national regulations. Permits authorising discharges to water are issued by the Council under Section 87(e) of the RMA.

Air discharge permits

Section 15(1)(c) of the RMA stipulates that no person may discharge any contaminant from any industrial or trade premises into air, unless the activity is expressly allowed for by a resource consent, a rule in a regional plan, or by national regulations. Permits authorising discharges to air are issued by the Council under Section 87(e) of the RMA.

Discharges of wastes to land

Sections 15(1)(b) and (d) of the RMA stipulate that no person may discharge any contaminant onto land if it may then enter water, or from any industrial or trade premises onto land under any circumstances, unless the activity is expressly allowed for by a resource consent, a rule in a regional plan, or by national regulations. Permits authorising the discharge of wastes to land are issued by the Council under Section 87(e) of the RMA.

Land use permits

Section 13(1)(a) of the RMA stipulates that no person may in relation to the bed of any lake or river use, erect, reconstruct, place, alter, extend, remove, or demolish any structure or part of any structure in, on, under, or over the bed, unless the activity is expressly allowed for by a resource consent, a rule in a regional plan, or by national regulations. Land use permits are issued by the Council under Section 87(a) of the RMA.

Coastal permits

Section 12(1)(b) of the RMA stipulates that no person may erect, reconstruct, place, alter, extend, remove, or demolish any structure that is fixed in, on, under, or over any foreshore or seabed, unless the activity is expressly allowed for by a resource consent, a rule in a regional plan, or by national regulations. Coastal permits are issued by the Council under Section 87(c) of the RMA.

Discharge Permit

Pursuant to the Resource Management Act 1991
a resource consent is hereby granted by the
Taranaki Regional Council

Name of Consent Holder:	Greymouth Petroleum Turangi Limited PO Box 3394 Fitzroy New Plymouth 4341
Decision Date:	9 June 2016
Commencement Date:	9 June 2016

Conditions of Consent

Consent Granted:	To discharge water based hydraulic fracturing fluids into land at depths greater than 3,410 mTVDss beneath the Turangi-B wellsite
Expiry Date:	1 June 2033
Review Date(s):	June annually and in accordance with special condition 20
Site Location:	Turangi-B wellsite, 42 Turangi Road Upper, Tikorangi (Property owner: RJ Topless)
Grid Reference (NZTM)	1713599E-5682493N
Catchment:	Parahaki

For General, Standard and Special conditions
pertaining to this consent please see reverse side of this document

General condition

a. The consent holder shall pay to the Taranaki Regional Council all the administration, monitoring and supervision costs of this consent, fixed in accordance with section 36 of the Resource Management Act 1991.

Special conditions

1. The discharge point shall be deeper than 3,410 mTVDss.

   Note: mTVDss = metres true vertical depth subsea, i.e., the true vertical depth in metres below mean sea level

2. There shall be no discharge of hydraulic fracturing fluids after 1 June 2028.

3. If the GeoNet seismic monitoring network records a seismic event higher than a Modified Mercalli intensity of magnitude 3.0 within 5 km of the geographical position (in 3 dimensions) of any hydraulic fracturing discharge, then:

   (a) if a hydraulic fracturing discharge is currently being undertaken it shall cease immediately and not recommence; or

   (b) if a hydraulic fracturing discharge has occurred within the previous 72 hours no further hydraulic fracturing discharges shall occur.

4. Following the occurrence of any seismic event described in special condition 3 the consent holder shall investigate and report to the Chief Executive, Taranaki Regional Council on the likelihood of the seismic event being induced by the exercise of this consent. Hydraulic fracturing discharges may only then continue once the Chief Executive, Taranaki Regional Council has considered the report and concluded that the environmental risk of recommencing hydraulic fracturing is acceptable and has advised the consent holder accordingly.

5. The consent holder shall ensure that the exercise of this consent does not result in contaminants reaching any useable fresh water (groundwater or surface water). Usable fresh groundwater is defined as any groundwater having a Total Dissolved Solids concentration of less than 1,000 mg/l.

6. The consent holder shall undertake a programme of sampling and testing that monitors the effects of the exercise of this consent on fresh water resources to assess compliance with condition 5 (the 'Monitoring Programme'). The Monitoring Programme shall be certified by the Chief Executive, Taranaki Regional Council ('the Chief Executive'), before this consent is exercised, and shall include:

   (a) the location of the discharge point(s);
   (b) the location of sampling sites; and
   (c) sampling frequency with reference to a hydraulic fracturing programme.

7. Depending on the suitability of existing bores within 500 metres of the wellsite for obtaining a representative groundwater sample, it may be necessary for the Monitoring Programme to include installation of, and sampling from, at least one monitoring bore. The bore(s) would be of a depth, location and design determined after consultation with the Chief Executive, Taranaki Regional Council and installed in accordance with NZS 4411:2001.

8. All water samples taken for monitoring purposes shall be taken in accordance with recognised field procedures and analysed for:

   (a) pH;
   (b) conductivity;
   (c) total dissolved solids;
   (d) major ions (Ca, Mg, K, Na, total alkalinity, bromide, chloride, nitrate-nitrogen, and sulphate);
   (e) trace metals (barium, copper, iron, manganese, nickel, and zinc);
   (f) total petroleum hydrocarbons;
   (g) formaldehyde;
   (h) dissolved methane and ethane gas;
   (i) methanol;
   (j) glycols;
   (k) benzene, toluene, ethylbenzene, and xylenes (BTEX); and
   (l) carbon-13 composition of any dissolved methane gas discovered (^{13}C-CH_4).

   Note: *The samples required, under conditions of this consent could be taken and analysed by the Taranaki Regional Council or other contracted party on behalf of the consent holder.*

9. All sampling and analysis shall be undertaken in accordance with a *Sampling and Analysis Plan*, which shall be submitted to the Chief Executive for review and certification before the first sampling is undertaken. The plan shall specify the use of standard protocols recognised to constitute good professional practice including quality control and assurance. An International Accreditation New Zealand (IANZ) accredited laboratory shall be used for all sample analysis. Results shall be provided to the Chief Executive within 30 days of sampling and shall include supporting quality control and assurance information. These results will be used to assess compliance with condition 5.

   Note: *The Sampling and Analysis Plan may be combined with the Monitoring Programme required by* condition 6.

10. The consent holder shall undertake well and equipment pressure testing prior to any hydraulic fracture programme on a given well to ensure any discharge will not affect the integrity of the well and hydraulic fracturing equipment.

11. Any hydraulic fracture discharge shall only occur after the consent holder has provided a comprehensive 'Pre-fracturing Discharge Report' to the Chief Executive. The report shall be provided at least 14 days before the discharge is proposed to commence and shall detail the hydraulic fracturing programme proposed, including as a minimum:

    (a) the specific well in which each discharge is to occur, the intended fracture interval(s) ('fracture interval' is the discrete subsurface zone to receive a hydraulic fracture treatment), and the duration of the hydraulic fracturing programme;
    (b) the number of discharges proposed and the geographical position (i.e. depth and lateral position) of each intended discharge point;
    (c) the total volume of fracture fluid planned to be pumped down the well, including mini-fracture treatments, and their intended composition, including a list of all contaminants and Material Safety Data Sheets for all the chemicals to be used;
    (d) the monitoring techniques to be used to determine the fate of discharged material;
    (e) the results of the reviews required by condition 17;
    (f) results of modelling showing an assessment of the likely extent and dimensions of the fractures that will be generated by the discharge;
    (g) the preventative and mitigation measures to be in place to ensure the discharge does not cause adverse environmental effects and complies with condition 5;
    (h) the extent and permeability characteristics of the geology above the discharge point to the surface;
    (i) any identified faults within the modelled fracture length plus a margin of 50%, and the potential for adverse environmental effects due to the presence of the identified faults;
    (j) the burst pressure of the well casing and the anticipated maximum well and discharge pressures and the duration of the pressures; and
    (k) details of the disposal of any returned fluids, including any consents that are relied on to authorise the disposal; and
    (l) details why the contaminants in the discharge and the monitoring techniques used comply with condition 17.

    Note: For the avoidance of doubt, the information provided with a resource consent application would usually be sufficient to constitute a 'Pre-fracturing Discharge Report' for any imminent hydraulic fracturing discharge. The Pre-fracturing Discharge Report provided for any later discharge may refer to the resource consent application or earlier Pre-fracturing Discharge Reports noting any differences.

12. The consent holder shall notify the Taranaki Regional Council of the date that each discharge is intended to commence by emailing worknotification@trc.govt.nz. Notification also shall identify the 'Pre-fracturing Discharge Report', required by condition 11, which details the discharge and be given no less than 3 days before the intended discharge date. If any discharge occurs more than 30 days after the notification date, additional notification as specified in this condition is required.

13. Subject to condition 14, within 90 days of any commencement date as advised under condition 12, the consent holder shall submit a comprehensive 'Post-fracturing Discharge Report' to the Chief Executive. The report shall, as a minimum, contain:

    (a) date and time of discharge;
    (b) confirmation of the interval(s) where fracturing occurred for that programme, and the geographical position (i.e., depth and lateral position) of the discharge point for each fracture interval;
    (c) the contaminant volumes and composition of fluid discharged into each fracture interval;
    (d) the volume of return fluids from each fracture interval;
    (e) an analysis for the constituents set out in conditions 8(a) to 8(k), in a return fluid sample taken within the first two hours of flow back, for each fracture interval if flowed back individually, or for the well if flowed back with all intervals comingled;
    (f) an estimate of the volume of fluids (and proppant) remaining underground;
    (g) the volume of water produced with the hydrocarbons (produced water) over the period beginning at the start of the hydraulic fracturing programme and ending 30 days after the programme is completed or after that period of production;
    (h) an assessment of the extent and dimensions of the fractures that were generated by the discharge, based on modelling undertaken after the discharge has occurred and other diagnostic techniques, including production analysis, available to determine fracture length, height and containment;
    (i) the results of pressure testing required by condition 10 and the top-hole pressure (psi), slurry rate (bpm), surface proppant concentration (lb/gal), bottom hole proppant concentration (lb/gal), and calculated bottom hole pressure (psi), as well as predicted values for each of these parameters; prior to, during and after each hydraulic fracture treatment;
    (j) details of the disposal of any returned fluids, including any consents that are relied on to authorise the disposal;
    (k) details of any incidents where hydraulic fracture fluid is unable to pass through the well perforations (screen outs) that occurred, their likely cause and implications for compliance with conditions 1 and 5; and
    (l) results of the monitoring referred to in condition 11 (d); and
    (m) an assessment of the effectiveness of the mitigation measures in place with specific reference to those described in the application for this consent.

14. On occasions, including for programs involving multiple hydraulic fracturing discharges, more than one 'Post-fracturing discharge report' may be required in order to meet the 90-day deadline from commencement required by condition 13. In these situations the consent holder shall submit an 'Interim Post-fracturing Discharge Report', which includes all the information that is available, to the Chief Executive within 90 days and a final Post-fracturing report as soon as practicable but within 90 days of the interim report.

15. The reports described in conditions 11 and 13 shall be emailed to consents@trc.govt.nz with a reference to the number of this consent.

16. The consent holder shall provide access to a location where the Taranaki Regional Council officers can obtain a sample of the hydraulic fracturing fluids and the return fluids.

17. The consent holder shall at all times adopt the best practicable option, as defined in section 2 of the Resource Management Act 1991, to prevent or minimize any actual or likely adverse effect of the activity on the environment by, as a minimum, ensuring that:

    (a) the discharge is contained within the fracture interval;
    (b) regular reviews of monitoring techniques used to ensure the discharge does not cause adverse environmental effects are undertaken;
    (c) regular reviews are undertaken of the preventative and mitigation measures adopted to ensure the discharge does not cause adverse environmental effects; and
    (d) regular reviews of the chemicals used are undertaken with a view to reducing the toxicity of the chemicals used.

18. The fracture fluid shall be comprised of no less than 95% water and proppant by volume.

19. This consent shall lapse on 30 June 2021, unless the consent is given effect to before the end of that period or the Taranaki Regional Council fixes a longer period pursuant to section 125(1)(b) of the Resource Management Act 1991.

20. The Taranaki Regional Council may review any or all of the conditions of this consent by giving notice of review:

    a) during the month of June each year, and/or
    b) within 30 days of receiving any investigation and report in accordance with special condition 4 above;

    for the purposes of:

    (a) ensuring that the conditions are adequate to deal with any significant adverse effects on the environment arising from the exercise of this consent, which were either not foreseen at the time the application was considered or which it was not appropriate to deal with at the time; and/or

    (b) further specifying the best practicable option as required by condition 17; and/or

    (c) ensuring hydraulic fracturing operations appropriately take into account any best practice guidance published by a recognised industry association or environmental regulator.

Signed at Stratford on 9 June 2016

For and on behalf of
Taranaki Regional Council

_____

A D McLay
Director - Resource Management

Appendix II

Categories used to evaluate environmental and administrative performance

Categories used to evaluate environmental and administrative performance

Environmental performance is concerned with actual or likely effects on the receiving environment from the activities during the monitoring year. Administrative performance is concerned with the Company's approach to demonstrating consent compliance in site operations and management including the timely provision of information to Council (such as contingency plans and water take data) in accordance with consent conditions.

Events that were beyond the control of the consent holder and unforeseeable (that is a defence under the provisions of the RMA can be established) may be excluded with regard to the performance rating applied. For example loss of data due to a flood destroying deployed field equipment.

The categories used by the Council for this monitoring period, and their interpretation, are as follows:

Environmental Performance

High: No or inconsequential (short-term duration, less than minor in severity) breaches of consent or regional plan parameters resulting from the activity; no adverse effects of significance noted or likely in the receiving environment. The Council did not record any verified unauthorised incidents involving environmental impacts and was not obliged to issue any abatement notices or infringement notices in relation to such impacts.

Good: Likely or actual adverse effects of activities on the receiving environment were negligible or minor at most. There were some such issues noted during monitoring, from self-reports, or during investigations of incidents reported to the Council by a third party but these items were not critical, and follow-up inspections showed they have been dealt with. These minor issues were resolved positively, co-operatively, and quickly. The Council was not obliged to issue any abatement notices or infringement notices in relation to the minor non-compliant effects; however abatement notices may have been issued to mitigate an identified potential for an environmental effect to occur.

For example:

- High suspended solid values recorded in discharge samples, however the discharge was to land or to receiving waters that were in high flow at the time;
- Strong odour beyond boundary but no residential properties or other recipient nearby.

Improvement required: Likely or actual adverse effects of activities on the receiving environment were more than minor, but not substantial. There were some issues noted during monitoring, from self-reports, or during investigations of incidents reported to the Council by a third party. Cumulative adverse effects of a persistent minor non-compliant activity could elevate a minor issue to this level. Abatement notices and infringement notices may have been issued in respect of effects.

Poor: Likely or actual adverse effects of activities on the receiving environment were significant. There were some items noted during monitoring, from self-reports, or during investigations of incidents reported to the Council by a third party. Cumulative adverse effects of a persistent moderate non-compliant activity could elevate an 'improvement required' issue to this level. Typically there were grounds for either a prosecution or an infringement notice in respect of effects.

Administrative performance

High: The administrative requirements of the resource consents were met, or any failure to do this had trivial consequences and were addressed promptly and co-operatively.

Good: Perhaps some administrative requirements of the resource consents were not met at a particular time, however this was addressed without repeated interventions from the Council staff. Alternatively adequate reason was provided for matters such as the no or late provision of information, interpretation of 'best practical option' for avoiding potential effects, etc.

Improvement required: Repeated interventions to meet the administrative requirements of the resource consents were made by Council staff. These matters took some time to resolve, or remained unresolved at the end of the period under review. The Council may have issued an abatement notice to attain compliance.

Poor: Material failings to meet the administrative requirements of the resource consents. Significant intervention by the Council was required. Typically there were grounds for an infringement notice.

Appendix III

Certificates of analysis (groundwater)

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

Client:	Taranaki Regional Council	**Lab No:**	2812323 SPv2
Contact:	Jane Harvey C/- Taranaki Regional Council Private Bag 713 Stratford 4352	**Date Received:**	24-Dec-2021
		Date Reported:	11-Jan-2022
		Quote No:	47915
		Order No:	4500002828
		Client Reference:	#7296- GPL Turangi-B 3 month PF+ Pre-frac Dec 2021
		Submitted By:	Sarah Larkin

Sample Type: Aqueous

Sample Name:		TRC214610 (GND3061) 23-Dec-2021 9:45 am				
Lab Number:		2812323.1				
Individual Tests						
Sum of Anions	meq/L	3.5	-	-	-	-
Sum of Cations	meq/L	3.6	-	-	-	-
pH	pH Units	7.1	-	-	-	-
Total Alkalinity	g/m³ as $CaCO_3$	149	-	-	-	-
Bicarbonate	g/m³ at 25°C	182	-	-	-	-
Total Hardness	g/m³ as $CaCO_3$	112	-	-	-	-
Electrical Conductivity (EC)	mS/m	32.7	-	-	-	-
Total Dissolved Solids (TDS)	g/m³	240	-	-	-	-
Sample Temperature*†	°C	17.0	-	-	-	-
Dissolved Barium	g/m³	0.047	-	-	-	-
Dissolved Calcium	g/m³	26	-	-	-	-
Dissolved Copper	g/m³	< 0.0005	-	-	-	-
Dissolved Iron	g/m³	8.3	-	-	-	-
Dissolved Magnesium	g/m³	11.4	-	-	-	-
Dissolved Manganese	g/m³	0.37	-	-	-	-
Dissolved Mercury	g/m³	< 0.00008	-	-	-	-
Dissolved Nickel	g/m³	< 0.0005	-	-	-	-
Dissolved Potassium	g/m³	5.5	-	-	-	-
Dissolved Sodium	g/m³	21	-	-	-	-
Dissolved Zinc	g/m³	0.0113	-	-	-	-
Bromide	g/m³	0.09	-	-	-	-
Chloride	g/m³	17.1	-	-	-	-
Nitrite-N	g/m³	< 0.02 #1	-	-	-	-
Nitrate-N	g/m³	< 0.02	-	-	-	-
Nitrate-N + Nitrite-N	g/m³	< 0.02 #1	-	-	-	-
Sulphate	g/m³	< 0.5	-	-	-	-
Ethylene Glycol in Water*						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water*						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*						
Methanol*	g/m³	< 2	-	-	-	-

Sample Type: Aqueous						
Sample Name:		TRC214610 (GND3061) 23-Dec-2021 9:45 am				
Lab Number:		2812323.1				
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.0010	-	-	-	-
Toluene	g/m³	< 0.0010	-	-	-	-
Ethylbenzene	g/m³	< 0.0010	-	-	-	-
m&p-Xylene	g/m³	< 0.002	-	-	-	-
o-Xylene	g/m³	< 0.0010	-	-	-	-
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde	g/m³	< 0.02	-	-	-	-
Gases in groundwater						
Ethane	g/m³	< 0.003	-	-	-	-
Ethylene	g/m³	< 0.004	-	-	-	-
Methane	g/m³	11.2	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	< 0.10	-	-	-	-
C10 - C14	g/m³	< 0.2	-	-	-	-
C15 - C36	g/m³	< 0.4	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	< 0.7	-	-	-	-

Analyst's Comments

† Customer supplied data. Please note: Hill Laboratories cannot be held responsible for the validity of this customer supplied data, or any subsequent calculations that rely on this information.

#1 Severe matrix interferences required that a dilution be performed prior to analysis, resulting in a detection limit higher than that normally achieved for the NOxN /NO2N analysis.

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Individual Tests			
Filtration, Unpreserved	Sample filtration through 0.45µm membrane filter.	-	1
Total anions for anion/cation balance check	Calculation: sum of anions as mEquiv/L calculated from Alkalinity (bicarbonate), Chloride and Sulphate. Nitrate-N, Nitrite-N. Fluoride, Dissolved Reactive Phosphorus and Cyanide also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.07 meq/L	1
Total cations for anion/cation balance check	Sum of cations as mEquiv/L calculated from Sodium, Potassium, Calcium and Magnesium. Iron, Manganese, Aluminium, Zinc, Copper, Lithium, Total Ammoniacal-N and pH (H^+) also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.05 meq/L	1
pH	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	1
Total Alkalinity	Titration to pH 4.5 (M-alkalinity), autotitrator. APHA 2320 B (modified for Alkalinity <20) 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Bicarbonate	Calculation: from alkalinity and pH, valid where TDS is not >500 mg/L and alkalinity is almost entirely due to hydroxides, carbonates or bicarbonates. APHA 4500-CO_2 D 23rd ed. 2017.	1.0 g/m³ at 25°C	1
Total Hardness	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Electrical Conductivity (EC)	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	1
Total Dissolved Solids (TDS)	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	10 g/m³	1

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Sample Temperature*	Temperature of the sample at the time of sampling, supplied by customer.	0.1 °C	1
Dissolved Barium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.005 g/m³	1
Dissolved Calcium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Copper	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Iron	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Magnesium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Manganese	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Mercury	0.45µm filtration, bromine oxidation followed by atomic fluorescence. US EPA Method 245.7, Feb 2005.	0.00008 g/m³	1
Dissolved Nickel	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Potassium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Sodium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Zinc	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	1
Bromide	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.05 g/m³	1
Chloride	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Nitrite-N	Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	1
Nitrate-N + Nitrite-N	Total oxidised nitrogen. Automated cadmium reduction, flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Sulphate	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Ethylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Methanol in Water - Aqueous Solvents*	GC-FID analysis. In-house.	1.0 g/m³	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis. In-house based on US EPA 8260 and 5021.	0.0010 - 0.002 g/m³	1
Formaldehyde in Water by DNPH & LCMSMS	Derivatisation, SPE extraction, LC-MS/MS analysis. In-house based on US EPA 8315A.	0.02 g/m³	1
Gases in groundwater	Headspace GC-FID analysis. In-house.	0.002 - 0.003 g/m³	1
Total Petroleum Hydrocarbons in Water			
C7 - C9	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.10 g/m³	1
C10 - C14	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.2 g/m³	1
C15 - C36	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.4 g/m³	1
Total hydrocarbons (C7 - C36)	Calculation: Sum of carbon bands from C7 to C36. In-house based on US EPA 8015.	0.7 g/m³	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Testing was completed between 24-Dec-2021 and 11-Jan-2022. For completion dates of individual analyses please contact the laboratory.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

Client:	Taranaki Regional Council	**Lab No:**	2328631 SPv1
Contact:	Jane Harvey	**Date Received:**	25-Feb-2020
	C/- Taranaki Regional Council	**Date Reported:**	04-Mar-2020
	Private Bag 713	**Quote No:**	47915
	Stratford 4352	**Order No:**	72831
		Client Reference:	#5688 - Greymouth Turangi B 3 month PF GW
		Submitted By:	Sarah Larkin

Sample Type: Aqueous

		Sample Name:	TRC200746 (GND3061) 24-Feb-2020 1:20 pm				
		Lab Number:	2328631.1				
Individual Tests							
Sum of Anions	meq/L		4.1	-	-	-	-
Sum of Cations	meq/L		4.0	-	-	-	-
pH	pH Units		7.4	-	-	-	-
Total Alkalinity	g/m³ as $CaCO_3$		180	-	-	-	-
Bicarbonate	g/m³ at 25°C		220	-	-	-	-
Total Hardness	g/m³ as $CaCO_3$		117	-	-	-	-
Electrical Conductivity (EC)	mS/m		37.9	-	-	-	-
Total Dissolved Solids (TDS)	g/m³		250	-	-	-	-
Sample Temperature*	°C		18.6	-	-	-	-
Dissolved Barium	g/m³		0.029	-	-	-	-
Dissolved Calcium	g/m³		27	-	-	-	-
Dissolved Copper	g/m³		< 0.0005	-	-	-	-
Dissolved Iron	g/m³		3.7	-	-	-	-
Dissolved Magnesium	g/m³		11.9	-	-	-	-
Dissolved Manganese	g/m³		0.28	-	-	-	-
Dissolved Mercury	g/m³		< 0.00008	-	-	-	-
Dissolved Nickel	g/m³		0.0006	-	-	-	-
Dissolved Potassium	g/m³		8.0	-	-	-	-
Dissolved Sodium	g/m³		31	-	-	-	-
Dissolved Zinc	g/m³		0.0027	-	-	-	-
Bromide	g/m³		0.09	-	-	-	-
Chloride	g/m³		16.4	-	-	-	-
Nitrite-N	g/m³		0.003	-	-	-	-
Nitrate-N	g/m³		0.030	-	-	-	-
Nitrate-N + Nitrite-N	g/m³		0.034	-	-	-	-
Sulphate	g/m³		< 0.5	-	-	-	-
Ethylene Glycol in Water*							
Ethylene glycol*	g/m³		< 4	-	-	-	-
Propylene Glycol in Water*							
Propylene glycol*	g/m³		< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*							
Methanol*	g/m³		< 2	-	-	-	-

Sample Type: Aqueous						
Sample Name:		TRC200746 (GND3061) 24-Feb-2020 1:20 pm				
Lab Number:		2328631.1				
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.0010	-	-	-	-
Toluene	g/m³	0.0014	-	-	-	-
Ethylbenzene	g/m³	< 0.0010	-	-	-	-
m&p-Xylene	g/m³	< 0.002	-	-	-	-
o-Xylene	g/m³	< 0.0010	-	-	-	-
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde	g/m³	< 0.02	-	-	-	-
Gases in groundwater						
Ethane	g/m³	< 0.003	-	-	-	-
Ethylene	g/m³	< 0.004	-	-	-	-
Methane	g/m³	8.2	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	< 0.10	-	-	-	-
C10 - C14	g/m³	< 0.2	-	-	-	-
C15 - C36	g/m³	< 0.4	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	< 0.7	-	-	-	-

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID.	1.0 g/m³	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629].	0.0010 - 0.002 g/m³	1
Formaldehyde in Water by DNPH & LCMSMS	DNPH derivatisation, extraction, LCMSMS.	0.02 g/m³	1
Gases in groundwater	Manual headspace creation and sub-sampling, GC-FID analysis.	0.002 - 0.003 g/m³	1
Total Petroleum Hydrocarbons in Water	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines.	0.10 - 0.7 g/m³	1
Filtration, Unpreserved	Sample filtration through 0.45µm membrane filter.	-	1
Total anions for anion/cation balance check	Calculation: sum of anions as mEquiv/L calculated from Alkalinity (bicarbonate), Chloride and Sulphate. Nitrate-N, Nitrite-N. Fluoride, Dissolved Reactive Phosphorus and Cyanide also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.07 meq/L	1
Total cations for anion/cation balance check	Sum of cations as mEquiv/L calculated from Sodium, Potassium, Calcium and Magnesium. Iron, Manganese, Aluminium, Zinc, Copper, Lithium, Total Ammoniacal-N and pH (H^+) also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.05 meq/L	1
pH	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	1
Total Alkalinity	Titration to pH 4.5 (M-alkalinity), autotitrator. APHA 2320 B (modified for Alkalinity <20) 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Bicarbonate	Calculation: from alkalinity and pH, valid where TDS is not >500 mg/L and alkalinity is almost entirely due to hydroxides, carbonates or bicarbonates. APHA 4500-CO_2 D 23rd ed. 2017.	1.0 g/m³ at 25°C	1
Total Hardness	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Electrical Conductivity (EC)	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	1
Total Dissolved Solids (TDS)	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	10 g/m³	1
Sample Temperature*	Supplied by customer, otherwise 20°C.	0.1 °C	1
Dissolved Barium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.005 g/m³	1
Dissolved Calcium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Copper	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Iron	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Magnesium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Manganese	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Mercury	0.45µm filtration, bromine oxidation followed by atomic fluorescence. US EPA Method 245.7, Feb 2005.	0.00008 g/m³	1
Dissolved Nickel	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Potassium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Sodium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Zinc	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	1
Bromide	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.05 g/m³	1
Chloride	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Nitrite-N	Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	1
Nitrate-N + Nitrite-N	Total oxidised nitrogen. Automated cadmium reduction, flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Sulphate	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

This certificate of analysis must not be reproduced, except in full, without the written consent of the signatory.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

Client:	Taranaki Regional Council	**Lab No:**	2539666 SPv1
Contact:	Jane Harvey C/- Taranaki Regional Council Private Bag 713 Stratford 4352	**Date Received:**	26-Feb-2021
		Date Reported:	05-Mar-2021
		Quote No:	47915
		Order No:	72831
		Client Reference:	#6601 - Greymouth Turangi-B 1yr PF/3 mnth PF/Pre-Frac
		Submitted By:	Kelby Clements

Sample Type: Aqueous

Sample Name:		TRC210881 (GND3061) 25-Feb-2021 11:05 am				
Lab Number:		2539666.1				
Individual Tests						
Sum of Anions	meq/L	3.8	-	-	-	-
Sum of Cations	meq/L	3.7	-	-	-	-
pH	pH Units	7.4	-	-	-	-
Total Alkalinity	g/m³ as $CaCO_3$	167	-	-	-	-
Bicarbonate	g/m³ at 25°C	200	-	-	-	-
Total Hardness	g/m³ as $CaCO_3$	108	-	-	-	-
Electrical Conductivity (EC)	mS/m	36.0	-	-	-	-
Total Dissolved Solids (TDS)	g/m³	220	-	-	-	-
Sample Temperature*†	°C	17.1	-	-	-	-
Dissolved Barium	g/m³	0.033	-	-	-	-
Dissolved Calcium	g/m³	24	-	-	-	-
Dissolved Copper	g/m³	< 0.0005	-	-	-	-
Dissolved Iron	g/m³	4.5	-	-	-	-
Dissolved Magnesium	g/m³	12.0	-	-	-	-
Dissolved Manganese	g/m³	0.23	-	-	-	-
Dissolved Mercury	g/m³	< 0.00008	-	-	-	-
Dissolved Nickel	g/m³	< 0.0005	-	-	-	-
Dissolved Potassium	g/m³	7.0	-	-	-	-
Dissolved Sodium	g/m³	28	-	-	-	-
Dissolved Zinc	g/m³	0.0052	-	-	-	-
Bromide	g/m³	0.06	-	-	-	-
Chloride	g/m³	16.3	-	-	-	-
Nitrite-N	g/m³	0.004	-	-	-	-
Nitrate-N	g/m³	0.012	-	-	-	-
Nitrate-N + Nitrite-N	g/m³	0.016	-	-	-	-
Sulphate	g/m³	< 0.5	-	-	-	-
Ethylene Glycol in Water*						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water*						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*						
Methanol*	g/m³	< 4	-	-	-	-

Sample Type: Aqueous						
Sample Name:		TRC210881 (GND3061) 25-Feb-2021 11:05 am				
Lab Number:		2539666.1				
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.0010	-	-	-	-
Toluene	g/m³	< 0.0010	-	-	-	-
Ethylbenzene	g/m³	< 0.0010	-	-	-	-
m&p-Xylene	g/m³	< 0.002	-	-	-	-
o-Xylene	g/m³	< 0.0010	-	-	-	-
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde	g/m³	< 0.02	-	-	-	-
Gases in groundwater						
Ethane	g/m³	< 0.003	-	-	-	-
Ethylene	g/m³	< 0.004	-	-	-	-
Methane	g/m³	8.3	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	< 0.10	-	-	-	-
C10 - C14	g/m³	< 0.2	-	-	-	-
C15 - C36	g/m³	< 0.4	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	< 0.7	-	-	-	-

Analyst's Comments

† Customer supplied data. Please note: Hill Laboratories cannot be held responsible for the validity of this customer supplied data, or any subsequent calculations that rely on this information.

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Individual Tests			
Filtration, Unpreserved	Sample filtration through 0.45µm membrane filter.	-	1
Total anions for anion/cation balance check	Calculation: sum of anions as mEquiv/L calculated from Alkalinity (bicarbonate), Chloride and Sulphate. Nitrate-N, Nitrite-N. Fluoride, Dissolved Reactive Phosphorus and Cyanide also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.07 meq/L	1
Total cations for anion/cation balance check	Sum of cations as mEquiv/L calculated from Sodium, Potassium, Calcium and Magnesium. Iron, Manganese, Aluminium, Zinc, Copper, Lithium, Total Ammoniacal-N and pH (H^+) also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.05 meq/L	1
pH	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	1
Total Alkalinity	Titration to pH 4.5 (M-alkalinity), autotitrator. APHA 2320 B (modified for Alkalinity <20) 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Bicarbonate	Calculation: from alkalinity and pH, valid where TDS is not >500 mg/L and alkalinity is almost entirely due to hydroxides, carbonates or bicarbonates. APHA 4500-CO_2 D 23rd ed. 2017.	1.0 g/m³ at 25°C	1
Total Hardness	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Electrical Conductivity (EC)	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	1
Total Dissolved Solids (TDS)	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	10 g/m³	1
Sample Temperature*	Temperature of the sample at the time of sampling, supplied by customer.	0.1 °C	1

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Dissolved Barium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.005 g/m³	1
Dissolved Calcium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Copper	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Iron	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Magnesium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Manganese	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Mercury	0.45µm filtration, bromine oxidation followed by atomic fluorescence. US EPA Method 245.7, Feb 2005.	0.00008 g/m³	1
Dissolved Nickel	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Potassium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Sodium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Zinc	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	1
Bromide	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.05 g/m³	1
Chloride	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Nitrite-N	Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	1
Nitrate-N + Nitrite-N	Total oxidised nitrogen. Automated cadmium reduction, flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Sulphate	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Ethylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Methanol in Water - Aqueous Solvents*	GC-FID analysis. In-house.	1.0 g/m³	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis. In-house based on US EPA 8260 and 5021.	0.0010 - 0.002 g/m³	1
Formaldehyde in Water by DNPH & LCMSMS	Derivatisation, SPE extraction, LC-MS/MS analysis. In-house based on US EPA 8315A.	0.02 g/m³	1
Gases in groundwater	Headspace GC-FID analysis. In-house.	0.002 - 0.003 g/m³	1
Total Petroleum Hydrocarbons in Water			
C7 - C9	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.10 g/m³	1
C10 - C14	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.2 g/m³	1
C15 - C36	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.4 g/m³	1
Total hydrocarbons (C7 - C36)	Calculation: Sum of carbon bands from C7 to C36. In-house based on US EPA 8015.	0.7 g/m³	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Testing was completed between 28-Feb-2021 and 05-Mar-2021. For completion dates of individual analyses please contact the laboratory.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

Client:	Taranaki Regional Council	**Lab No:**	2409527 SPv1
Contact:	Jane Harvey	**Date Received:**	30-Jul-2020
	C/- Taranaki Regional Council	**Date Reported:**	06-Aug-2020
	Private Bag 713	**Quote No:**	47915
	Stratford 4352	**Order No:**	72831
		Client Reference:	#5932 - Greymouth Turangi B Interim 3 month PF GW
		Submitted By:	Sarah Larkin

Sample Type: Aqueous

Sample Name:		TRC202067 (GND3061) 29-Jul-2020 11:00 am				
Lab Number:		2409527.1				
Individual Tests						
Sum of Anions	meq/L	3.6	-	-	-	-
Sum of Cations	meq/L	3.8	-	-	-	-
pH	pH Units	6.9	-	-	-	-
Total Alkalinity	g/m³ as $CaCO_3$	155	-	-	-	-
Bicarbonate	g/m³ at 25°C	189	-	-	-	-
Total Hardness	g/m³ as $CaCO_3$	122	-	-	-	-
Electrical Conductivity (EC)	mS/m	34.2	-	-	-	-
Total Dissolved Solids (TDS)	g/m³	250	-	-	-	-
Sample Temperature*†	°C	15.5	-	-	-	-
Dissolved Barium	g/m³	0.044	-	-	-	-
Dissolved Calcium	g/m³	30	-	-	-	-
Dissolved Copper	g/m³	< 0.0005	-	-	-	-
Dissolved Iron	g/m³	8.0	-	-	-	-
Dissolved Magnesium	g/m³	11.5	-	-	-	-
Dissolved Manganese	g/m³	0.45	-	-	-	-
Dissolved Mercury	g/m³	< 0.00008	-	-	-	-
Dissolved Nickel	g/m³	0.0006	-	-	-	-
Dissolved Potassium	g/m³	6.0	-	-	-	-
Dissolved Sodium	g/m³	21	-	-	-	-
Dissolved Zinc	g/m³	0.0130	-	-	-	-
Bromide	g/m³	< 0.05	-	-	-	-
Chloride	g/m³	16.2	-	-	-	-
Nitrite-N	g/m³	0.005 #1	-	-	-	-
Nitrate-N	g/m³	< 0.002	-	-	-	-
Nitrate-N + Nitrite-N	g/m³	0.003 #1	-	-	-	-
Sulphate	g/m³	0.5	-	-	-	-
Ethylene Glycol in Water*						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water*						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*						
Methanol*	g/m³	< 2	-	-	-	-

Sample Type: Aqueous						
Sample Name:		TRC202067 (GND3061) 29-Jul-2020 11:00 am				
Lab Number:		2409527.1				
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.0010	-	-	-	-
Toluene	g/m³	< 0.0010	-	-	-	-
Ethylbenzene	g/m³	< 0.0010	-	-	-	-
m&p-Xylene	g/m³	< 0.002	-	-	-	-
o-Xylene	g/m³	< 0.0010	-	-	-	-
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde	g/m³	0.02	-	-	-	-
Gases in groundwater						
Ethane	g/m³	< 0.003	-	-	-	-
Ethylene	g/m³	< 0.003	-	-	-	-
Methane	g/m³	7.7	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	< 0.10	-	-	-	-
C10 - C14	g/m³	< 0.2	-	-	-	-
C15 - C36	g/m³	< 0.4	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	< 0.7	-	-	-	-

Analyst's Comments

† Customer supplied data. Please note: Hill Laboratories cannot be held responsible for the validity of this customer supplied data, or any subsequent calculations that rely on this information.

#1 It has been noted that the result for Nitrite-N was greater than that for Nitrate-N + Nitrite-N, but within the analytical variation of these methods.

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

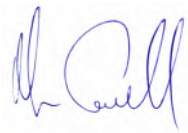
Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Individual Tests			
Filtration, Unpreserved	Sample filtration through 0.45µm membrane filter.	-	1
Total anions for anion/cation balance check	Calculation: sum of anions as mEquiv/L calculated from Alkalinity (bicarbonate), Chloride and Sulphate. Nitrate-N, Nitrite-N. Fluoride, Dissolved Reactive Phosphorus and Cyanide also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.07 meq/L	1
Total cations for anion/cation balance check	Sum of cations as mEquiv/L calculated from Sodium, Potassium, Calcium and Magnesium. Iron, Manganese, Aluminium, Zinc, Copper, Lithium, Total Ammoniacal-N and pH (H^+) also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.05 meq/L	1
pH	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	1
Total Alkalinity	Titration to pH 4.5 (M-alkalinity), autotitrator. APHA 2320 B (modified for Alkalinity <20) 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Bicarbonate	Calculation: from alkalinity and pH, valid where TDS is not >500 mg/L and alkalinity is almost entirely due to hydroxides, carbonates or bicarbonates. APHA 4500-CO_2 D 23rd ed. 2017.	1.0 g/m³ at 25°C	1
Total Hardness	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Electrical Conductivity (EC)	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	1
Total Dissolved Solids (TDS)	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	10 g/m³	1

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Sample Temperature*	Temperature of the sample at the time of sampling, supplied by customer.	0.1 °C	1
Dissolved Barium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.005 g/m³	1
Dissolved Calcium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Copper	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Iron	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Magnesium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Manganese	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Mercury	0.45µm filtration, bromine oxidation followed by atomic fluorescence. US EPA Method 245.7, Feb 2005.	0.00008 g/m³	1
Dissolved Nickel	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Potassium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Sodium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Zinc	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	1
Bromide	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.05 g/m³	1
Chloride	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Nitrite-N	Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	1
Nitrate-N + Nitrite-N	Total oxidised nitrogen. Automated cadmium reduction, flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Sulphate	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Ethylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Methanol in Water - Aqueous Solvents*	GC-FID analysis. In-house.	1.0 g/m³	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis. In-house based on US EPA 8260 and 5021.	0.0010 - 0.002 g/m³	1
Formaldehyde in Water by DNPH & LCMSMS	Derivatisation, SPE extraction, LC-MS/MS analysis. In-house based on US EPA 8315A.	0.02 g/m³	1
Gases in groundwater	Headspace GC-FID analysis. In-house.	0.002 - 0.003 g/m³	1
Total Petroleum Hydrocarbons in Water			
C7 - C9	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.10 g/m³	1
C10 - C14	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.2 g/m³	1
C15 - C36	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.4 g/m³	1
Total hydrocarbons (C7 - C36)	Calculation: Sum of carbon bands from C7 to C36. In-house based on US EPA 8015.	0.7 g/m³	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.



Martin Cowell - BSc
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

Client:	Taranaki Regional Council	**Lab No:**	2654099 SPv1
Contact:	Jane Harvey C/- Taranaki Regional Council Private Bag 713 Stratford 4352	**Date Received:**	10-Jul-2021
		Date Reported:	20-Jul-2021
		Quote No:	47915
		Order No:	4500002828
		Client Reference:	#6847- Turangi-B interim frac
		Submitted By:	Angela Collins

Sample Type: Aqueous

Sample Name:		TRC212297 (GND3061) 09-Jul-2021 11:20 am				
Lab Number:		2654099.1				
Individual Tests						
Sum of Anions	meq/L	3.7	-	-	-	-
Sum of Cations	meq/L	3.9	-	-	-	-
pH	pH Units	6.9	-	-	-	-
Total Alkalinity	g/m³ as $CaCO_3$	159	-	-	-	-
Bicarbonate	g/m³ at 25°C	194	-	-	-	-
Total Hardness	g/m³ as $CaCO_3$	125	-	-	-	-
Electrical Conductivity (EC)	mS/m	36.1	-	-	-	-
Total Dissolved Solids (TDS)	g/m³	240	-	-	-	-
Sample Temperature*†	°C	15.0	-	-	-	-
Dissolved Barium	g/m³	0.064	-	-	-	-
Dissolved Calcium	g/m³	30	-	-	-	-
Dissolved Copper	g/m³	< 0.0005	-	-	-	-
Dissolved Iron	g/m³	10.8	-	-	-	-
Dissolved Magnesium	g/m³	12.3	-	-	-	-
Dissolved Manganese	g/m³	0.48	-	-	-	-
Dissolved Mercury	g/m³	< 0.00008	-	-	-	-
Dissolved Nickel	g/m³	0.0005	-	-	-	-
Dissolved Potassium	g/m³	5.2	-	-	-	-
Dissolved Sodium	g/m³	21	-	-	-	-
Dissolved Zinc	g/m³	0.0044	-	-	-	-
Bromide	g/m³	0.09	-	-	-	-
Chloride	g/m³	16.4	-	-	-	-
Nitrite-N	g/m³	< 0.02 #1	-	-	-	-
Nitrate-N	g/m³	< 0.02	-	-	-	-
Nitrate-N + Nitrite-N	g/m³	< 0.02 #1	-	-	-	-
Sulphate	g/m³	1.0	-	-	-	-
Ethylene Glycol in Water*						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water*						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*						
Methanol*	g/m³	< 2	-	-	-	-

Sample Type: Aqueous						
Sample Name:		TRC212297 (GND3061) 09-Jul-2021 11:20 am				
Lab Number:		2654099.1				
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.0010	-	-	-	-
Toluene	g/m³	< 0.0010	-	-	-	-
Ethylbenzene	g/m³	< 0.0010	-	-	-	-
m&p-Xylene	g/m³	< 0.002	-	-	-	-
o-Xylene	g/m³	< 0.0010	-	-	-	-
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde	g/m³	0.02	-	-	-	-
Gases in groundwater						
Ethane	g/m³	< 0.003	-	-	-	-
Ethylene	g/m³	< 0.004	-	-	-	-
Methane	g/m³	14.2	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	< 0.10	-	-	-	-
C10 - C14	g/m³	< 0.2	-	-	-	-
C15 - C36	g/m³	< 0.4	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	< 0.7	-	-	-	-

Analyst's Comments

It has been noted that the spike for Methanol on sample 2654099.1, was run as part of our in-house QC procedure, had a lower than expected recovery at 44%. Therefore the result may be underestimated.

† Customer supplied data. Please note: Hill Laboratories cannot be held responsible for the validity of this customer supplied data, or any subsequent calculations that rely on this information.

#1 Due to the nature of this sample a dilution was performed prior to analysis, resulting in a detection limit higher than that normally achieved for the NO2N, NO3N and NOxN analysis.

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Individual Tests			
Filtration, Unpreserved	Sample filtration through 0.45µm membrane filter.	-	1
Total anions for anion/cation balance check	Calculation: sum of anions as mEquiv/L calculated from Alkalinity (bicarbonate), Chloride and Sulphate. Nitrate-N, Nitrite-N. Fluoride, Dissolved Reactive Phosphorus and Cyanide also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.07 meq/L	1
Total cations for anion/cation balance check	Sum of cations as mEquiv/L calculated from Sodium, Potassium, Calcium and Magnesium. Iron, Manganese, Aluminium, Zinc, Copper, Lithium, Total Ammoniacal-N and pH (H^+) also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.05 meq/L	1
pH	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	1
Total Alkalinity	Titration to pH 4.5 (M-alkalinity), autotitrator. APHA 2320 B (modified for Alkalinity <20) 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Bicarbonate	Calculation: from alkalinity and pH, valid where TDS is not >500 mg/L and alkalinity is almost entirely due to hydroxides, carbonates or bicarbonates. APHA 4500-CO_2 D 23rd ed. 2017.	1.0 g/m³ at 25°C	1
Total Hardness	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Electrical Conductivity (EC)	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	1

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Total Dissolved Solids (TDS)	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	10 g/m³	1
Sample Temperature*	Temperature of the sample at the time of sampling, supplied by customer.	0.1 °C	1
Dissolved Barium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.005 g/m³	1
Dissolved Calcium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Copper	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Iron	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Magnesium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Manganese	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Mercury	0.45µm filtration, bromine oxidation followed by atomic fluorescence. US EPA Method 245.7, Feb 2005.	0.00008 g/m³	1
Dissolved Nickel	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Potassium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Sodium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Zinc	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	1
Bromide	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.05 g/m³	1
Chloride	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Nitrite-N	Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	1
Nitrate-N + Nitrite-N	Total oxidised nitrogen. Automated cadmium reduction, flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Sulphate	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Ethylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Methanol in Water - Aqueous Solvents*	GC-FID analysis. In-house.	1.0 g/m³	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis. In-house based on US EPA 8260 and 5021.	0.0010 - 0.002 g/m³	1
Formaldehyde in Water by DNPH & LCMSMS	Derivatisation, SPE extraction, LC-MS/MS analysis. In-house based on US EPA 8315A.	0.02 g/m³	1
Gases in groundwater	Headspace GC-FID analysis. In-house.	0.002 - 0.003 g/m³	1
Total Petroleum Hydrocarbons in Water			
C7 - C9	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.10 g/m³	1
C10 - C14	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.2 g/m³	1
C15 - C36	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.4 g/m³	1
Total hydrocarbons (C7 - C36)	Calculation: Sum of carbon bands from C7 to C36. In-house based on US EPA 8015.	0.7 g/m³	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Testing was completed between 12-Jul-2021 and 20-Jul-2021. For completion dates of individual analyses please contact the laboratory.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

This certificate of analysis must not be reproduced, except in full, without the written consent of the signatory.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

Client:	Taranaki Regional Council	**Lab No:**	3013927 SPv1
Contact:	Jane Harvey C/- Taranaki Regional Council Private Bag 713 Stratford 4352	**Date Received:**	15-Jun-2022
		Date Reported:	22-Jun-2022
		Quote No:	47915
		Order No:	4500002828
		Client Reference:	#7622 - GPL Turangi-B 3 month PF + 1 Yr PF GW June 2022
		Submitted By:	Adele Bittner

Sample Type: Aqueous

Sample Name:		TRC226761 (GND3061) 14-Jun-2022 12:47 pm				
Lab Number:		3013927.1				
Individual Tests						
Sum of Anions	meq/L	3.3	-	-	-	-
Sum of Cations	meq/L	3.4	-	-	-	-
pH	pH Units	7.1	-	-	-	-
Total Alkalinity	g/m³ as $CaCO_3$	141	-	-	-	-
Bicarbonate	g/m³ at 25°C	172	-	-	-	-
Total Hardness	g/m³ as $CaCO_3$	105	-	-	-	-
Electrical Conductivity (EC)	mS/m	31.2	-	-	-	-
Total Dissolved Solids (TDS)	g/m³	260	-	-	-	-
Sample Temperature*†	°C	15.0	-	-	-	-
Dissolved Barium	g/m³	0.051	-	-	-	-
Dissolved Calcium	g/m³	24	-	-	-	-
Dissolved Copper	g/m³	< 0.0005	-	-	-	-
Dissolved Iron	g/m³	9.4	-	-	-	-
Dissolved Magnesium	g/m³	10.8	-	-	-	-
Dissolved Manganese	g/m³	0.38	-	-	-	-
Dissolved Mercury	g/m³	< 0.00008	-	-	-	-
Dissolved Nickel	g/m³	0.0006	-	-	-	-
Dissolved Potassium	g/m³	4.2	-	-	-	-
Dissolved Sodium	g/m³	18.8	-	-	-	-
Dissolved Zinc	g/m³	0.0116	-	-	-	-
Bromide	g/m³	0.07	-	-	-	-
Chloride	g/m³	16.6	-	-	-	-
Nitrite-N	g/m³	< 0.02 #1	-	-	-	-
Nitrate-N	g/m³	< 0.02	-	-	-	-
Nitrate-N + Nitrite-N	g/m³	< 0.02 #1	-	-	-	-
Sulphate	g/m³	< 0.5	-	-	-	-
Ethylene Glycol in Water*						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water*						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*						
Methanol*	g/m³	< 2	-	-	-	-

Sample Type: Aqueous						
Sample Name:		TRC226761 (GND3061) 14-Jun-2022 12:47 pm				
Lab Number:		3013927.1				
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.0010	-	-	-	-
Toluene	g/m³	0.0014	-	-	-	-
Ethylbenzene	g/m³	< 0.0010	-	-	-	-
m&p-Xylene	g/m³	< 0.002	-	-	-	-
o-Xylene	g/m³	< 0.0010	-	-	-	-
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde	g/m³	< 0.02	-	-	-	-
Gases in groundwater						
Ethane	g/m³	< 0.003	-	-	-	-
Ethylene	g/m³	< 0.004	-	-	-	-
Methane	g/m³	11.7	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	< 0.10	-	-	-	-
C10 - C14	g/m³	< 0.2	-	-	-	-
C15 - C36	g/m³	< 0.4	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	< 0.7	-	-	-	-

Analyst's Comments

† Customer supplied data. Please note: Hill Laboratories cannot be held responsible for the validity of this customer supplied data, or any subsequent calculations that rely on this information.

#1 Severe matrix interferences required that a dilution be performed prior to analysis of this sample, resulting in a detection limit higher than that normally achieved for the NO2N, NO3N and NOxN analysis.

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

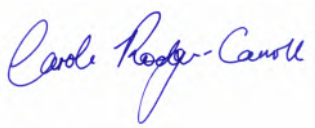
Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Individual Tests			
Filtration, Unpreserved	Sample filtration through 0.45µm membrane filter.	-	1
Total anions for anion/cation balance check	Calculation: sum of anions as mEquiv/L calculated from Alkalinity (bicarbonate), Chloride and Sulphate. Nitrate-N, Nitrite-N. Fluoride, Dissolved Reactive Phosphorus and Cyanide also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.07 meq/L	1
Total cations for anion/cation balance check	Sum of cations as mEquiv/L calculated from Sodium, Potassium, Calcium and Magnesium. Iron, Manganese, Aluminium, Zinc, Copper, Lithium, Total Ammoniacal-N and pH (H^+) also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.05 meq/L	1
pH	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	1
Total Alkalinity	Titration to pH 4.5 (M-alkalinity), autotitrator. APHA 2320 B (modified for Alkalinity <20) 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Bicarbonate	Calculation: from alkalinity and pH, valid where TDS is not >500 mg/L and alkalinity is almost entirely due to hydroxides, carbonates or bicarbonates. APHA 4500-CO_2 D 23rd ed. 2017.	1.0 g/m³ at 25°C	1
Total Hardness	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Electrical Conductivity (EC)	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	1
Total Dissolved Solids (TDS)	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	10 g/m³	1

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Sample Temperature*	Temperature of the sample at the time of sampling, supplied by customer.	0.1 °C	1
Dissolved Barium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.005 g/m³	1
Dissolved Calcium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Copper	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Iron	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Magnesium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Manganese	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Mercury	0.45µm filtration, bromine oxidation followed by atomic fluorescence. US EPA Method 245.7, Feb 2005.	0.00008 g/m³	1
Dissolved Nickel	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Potassium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Sodium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Zinc	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	1
Bromide	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.05 g/m³	1
Chloride	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Nitrite-N	Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	1
Nitrate-N + Nitrite-N	Total oxidised nitrogen. Automated cadmium reduction, flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Sulphate	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Ethylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Methanol in Water - Aqueous Solvents*	GC-FID analysis. In-house.	1.0 g/m³	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis. In-house based on US EPA 8260 and 5021.	0.0010 - 0.002 g/m³	1
Formaldehyde in Water by DNPH & LCMSMS	Derivatisation, SPE extraction, LC-MS/MS analysis. In-house based on US EPA 8315A.	0.02 g/m³	1
Gases in groundwater	Headspace GC-FID analysis. In-house.	0.002 - 0.003 g/m³	1
Total Petroleum Hydrocarbons in Water			
C7 - C9	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.10 g/m³	1
C10 - C14	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.2 g/m³	1
C15 - C36	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.4 g/m³	1
Total hydrocarbons (C7 - C36)	Calculation: Sum of carbon bands from C7 to C36. In-house based on US EPA 8015.	0.7 g/m³	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Testing was completed between 16-Jun-2022 and 22-Jun-2022. For completion dates of individual analyses please contact the laboratory.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.



Carole Rodgers-Carroll BA, NZCS
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

Client:	Taranaki Regional Council	**Lab No:**	2272692 SPv1
Contact:	Jane Harvey	**Date Received:**	09-Nov-2019
	C/- Taranaki Regional Council	**Date Reported:**	18-Nov-2019
	Private Bag 713	**Quote No:**	47915
	Stratford 4352	**Order No:**	72831
		Client Reference:	#5457 - Greymouth Petroleum Turangi B HF Pre Frac GW
		Submitted By:	Sarah Larkin

Sample Type: Aqueous

Sample Name:		TRC193990 GND3061 08-Nov-2019 12:15 pm				
Lab Number:		2272692.1				
Individual Tests						
Sum of Anions	meq/L	4.2	-	-	-	-
Sum of Cations	meq/L	4.5	-	-	-	-
pH	pH Units	7.5	-	-	-	-
Total Alkalinity	g/m³ as $CaCO_3$	182	-	-	-	-
Bicarbonate	g/m³ at 25°C	220	-	-	-	-
Total Hardness	g/m³ as $CaCO_3$	146	-	-	-	-
Electrical Conductivity (EC)	mS/m	40.8	-	-	-	-
Total Dissolved Solids (TDS)	g/m³	240	-	-	-	-
Sample Temperature*	°C	18.4	-	-	-	-
Dissolved Barium	g/m³	0.034	-	-	-	-
Dissolved Calcium	g/m³	36	-	-	-	-
Dissolved Copper	g/m³	< 0.0005	-	-	-	-
Dissolved Iron	g/m³	0.07	-	-	-	-
Dissolved Magnesium	g/m³	13.4	-	-	-	-
Dissolved Manganese	g/m³	0.25	-	-	-	-
Dissolved Mercury	g/m³	< 0.00008	-	-	-	-
Dissolved Nickel	g/m³	0.0011	-	-	-	-
Dissolved Potassium	g/m³	9.7	-	-	-	-
Dissolved Sodium	g/m³	30	-	-	-	-
Dissolved Zinc	g/m³	< 0.0010	-	-	-	-
Bromide	g/m³	0.11	-	-	-	-
Chloride	g/m³	18.3	-	-	-	-
Nitrite-N	g/m³	< 0.002	-	-	-	-
Nitrate-N	g/m³	0.003	-	-	-	-
Nitrate-N + Nitrite-N	g/m³	0.003	-	-	-	-
Sulphate	g/m³	4.9	-	-	-	-
Ethylene Glycol in Water						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents						
Methanol*	g/m³	< 2	-	-	-	-

Sample Type: Aqueous						
Sample Name:		TRC193990 GND3061 08-Nov-2019 12:15 pm				
Lab Number:		2272692.1				
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.0010	-	-	-	-
Toluene	g/m³	0.0023	-	-	-	-
Ethylbenzene	g/m³	< 0.0010	-	-	-	-
m&p-Xylene	g/m³	< 0.002	-	-	-	-
o-Xylene	g/m³	< 0.0010	-	-	-	-
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde	g/m³	< 0.02	-	-	-	-
Gases in groundwater						
Ethane	g/m³	< 0.003	-	-	-	-
Ethylene	g/m³	< 0.004	-	-	-	-
Methane	g/m³	7.5	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	< 0.10	-	-	-	-
C10 - C14	g/m³	< 0.2	-	-	-	-
C15 - C36	g/m³	< 0.4	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	< 0.7	-	-	-	-

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively clean matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	1
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m³	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m³	1
Formaldehyde in Water by DNPH & LCMSMS	DNPH derivatisation, extraction, LCMSMS	0.02 g/m³	1
Gases in groundwater	Manual headspace creation and sub-sampling, GC-FID analysis.	0.002 - 0.003 g/m³	1
Total Petroleum Hydrocarbons in Water	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m³	1
Filtration, Unpreserved	Sample filtration through 0.45µm membrane filter.	-	1
Total anions for anion/cation balance check	Calculation: sum of anions as mEquiv/L calculated from Alkalinity (bicarbonate), Chloride and Sulphate. Nitrate-N, Nitrite-N. Fluoride, Dissolved Reactive Phosphorus and Cyanide also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.07 meq/L	1
Total cations for anion/cation balance check	Sum of cations as mEquiv/L calculated from Sodium, Potassium, Calcium and Magnesium. Iron, Manganese, Aluminium, Zinc, Copper, Lithium, Total Ammoniacal-N and pH (H^+) also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.05 meq/L	1
pH	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	1
Total Alkalinity	Titration to pH 4.5 (M-alkalinity), autotitrator. APHA 2320 B (modified for Alkalinity <20) 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Bicarbonate	Calculation: from alkalinity and pH, valid where TDS is not >500 mg/L and alkalinity is almost entirely due to hydroxides, carbonates or bicarbonates. APHA 4500-CO_2 D 23rd ed. 2017.	1.0 g/m³ at 25°C	1
Total Hardness	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Electrical Conductivity (EC)	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	1

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Total Dissolved Solids (TDS)	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	10 g/m³	1
Sample Temperature*	Supplied by customer, otherwise 20°C.	0.1 °C	1
Filtration for dissolved metals analysis	Sample filtration through 0.45µm membrane filter and preservation with nitric acid. APHA 3030 B 23rd ed. 2017.	-	1
Dissolved Barium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.005 g/m³	1
Dissolved Calcium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Copper	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Iron	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Magnesium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Manganese	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Mercury	0.45µm filtration, bromine oxidation followed by atomic fluorescence. US EPA Method 245.7, Feb 2005.	0.00008 g/m³	1
Dissolved Nickel	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Potassium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Sodium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Zinc	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	1
Bromide	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.05 g/m³	1
Chloride	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Nitrite-N	Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	1
Nitrate-N + Nitrite-N	Total oxidised nitrogen. Automated cadmium reduction, flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Sulphate	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
C7 - C14	Solvent extraction, GC-FID analysis. US EPA 8015B/NZ OIEWG.	0.3 g/m³	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Samples are held at the laboratory after reporting for a length of time depending on the preservation used and the stability of the analytes being tested. Once the storage period is completed the samples are discarded unless otherwise advised by the client.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

Client:	Taranaki Regional Council	**Lab No:**	2470916 SPv1
Contact:	Jane Harvey	**Date Received:**	11-Nov-2020
	C/- Taranaki Regional Council	**Date Reported:**	19-Nov-2020
	Private Bag 713	**Quote No:**	47915
	Stratford 4352	**Order No:**	72831
		Client Reference:	#6354 - Turangi-B 1 year Post-frac GW
		Submitted By:	Grace Sommerville

Sample Type: Aqueous

Sample Name:		TRC203765 (GND3061) 10-Nov-2020 10:25 am				
Lab Number:		2470916.1				
Individual Tests						
Sum of Anions	meq/L	3.8	-	-	-	-
Sum of Cations	meq/L	3.8	-	-	-	-
pH	pH Units	7.1	-	-	-	-
Total Alkalinity	g/m³ as $CaCO_3$	166	-	-	-	-
Bicarbonate	g/m³ at 25°C	200	-	-	-	-
Total Hardness	g/m³ as $CaCO_3$	117	-	-	-	-
Electrical Conductivity (EC)	mS/m	35.7	-	-	-	-
Total Dissolved Solids (TDS)	g/m³	260	-	-	-	-
Sample Temperature*†	°C	17.1	-	-	-	-
Dissolved Barium	g/m³	0.044	-	-	-	-
Dissolved Calcium	g/m³	28	-	-	-	-
Dissolved Copper	g/m³	< 0.0005	-	-	-	-
Dissolved Iron	g/m³	9.5	-	-	-	-
Dissolved Magnesium	g/m³	11.6	-	-	-	-
Dissolved Manganese	g/m³	0.48	-	-	-	-
Dissolved Mercury	g/m³	< 0.00008	-	-	-	-
Dissolved Nickel	g/m³	< 0.0005	-	-	-	-
Dissolved Potassium	g/m³	6.4	-	-	-	-
Dissolved Sodium	g/m³	23	-	-	-	-
Dissolved Zinc	g/m³	0.0062	-	-	-	-
Bromide	g/m³	0.09	-	-	-	-
Chloride	g/m³	15.5	-	-	-	-
Nitrite-N	g/m³	< 0.02 #1	-	-	-	-
Nitrate-N	g/m³	< 0.02	-	-	-	-
Nitrate-N + Nitrite-N	g/m³	< 0.02 #1	-	-	-	-
Sulphate	g/m³	< 0.5	-	-	-	-
Ethylene Glycol in Water*						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water*						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*						
Methanol*	g/m³	< 2	-	-	-	-

Sample Type: Aqueous						
Sample Name:		TRC203765 (GND3061) 10-Nov-2020 10:25 am				
Lab Number:		2470916.1				
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.0010	-	-	-	-
Toluene	g/m³	< 0.0010	-	-	-	-
Ethylbenzene	g/m³	< 0.0010	-	-	-	-
m&p-Xylene	g/m³	< 0.002	-	-	-	-
o-Xylene	g/m³	< 0.0010	-	-	-	-
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde	g/m³	< 0.02	-	-	-	-
Gases in groundwater						
Ethane	g/m³	< 0.003	-	-	-	-
Ethylene	g/m³	< 0.004	-	-	-	-
Methane	g/m³	11.5	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	< 0.10	-	-	-	-
C10 - C14	g/m³	< 0.2	-	-	-	-
C15 - C36	g/m³	< 0.4	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	< 0.7	-	-	-	-

Analyst's Comments

† Customer supplied data. Please note: Hill Laboratories cannot be held responsible for the validity of this customer supplied data, or any subsequent calculations that rely on this information.

#1 Severe matrix interferences required that a dilution be performed prior to analysis of this sample, resulting in a detection limit higher than that normally achieved for the NO2N, NO3N and NOxN analysis.

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Individual Tests			
Filtration, Unpreserved	Sample filtration through 0.45µm membrane filter.	-	1
Total anions for anion/cation balance check	Calculation: sum of anions as mEquiv/L calculated from Alkalinity (bicarbonate), Chloride and Sulphate. Nitrate-N, Nitrite-N. Fluoride, Dissolved Reactive Phosphorus and Cyanide also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.07 meq/L	1
Total cations for anion/cation balance check	Sum of cations as mEquiv/L calculated from Sodium, Potassium, Calcium and Magnesium. Iron, Manganese, Aluminium, Zinc, Copper, Lithium, Total Ammoniacal-N and pH (H^+) also included in calculation if available. APHA 1030 E 23rd ed. 2017.	0.05 meq/L	1
pH	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	1
Total Alkalinity	Titration to pH 4.5 (M-alkalinity), autotitrator. APHA 2320 B (modified for Alkalinity <20) 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Bicarbonate	Calculation: from alkalinity and pH, valid where TDS is not >500 mg/L and alkalinity is almost entirely due to hydroxides, carbonates or bicarbonates. APHA 4500-CO_2 D 23rd ed. 2017.	1.0 g/m³ at 25°C	1
Total Hardness	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	1
Electrical Conductivity (EC)	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	1
Total Dissolved Solids (TDS)	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	10 g/m³	1

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Sample Temperature*	Temperature of the sample at the time of sampling, supplied by customer.	0.1 °C	1
Dissolved Barium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.005 g/m³	1
Dissolved Calcium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Copper	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Iron	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Magnesium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Manganese	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Mercury	0.45µm filtration, bromine oxidation followed by atomic fluorescence. US EPA Method 245.7, Feb 2005.	0.00008 g/m³	1
Dissolved Nickel	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0005 g/m³	1
Dissolved Potassium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.05 g/m³	1
Dissolved Sodium	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.02 g/m³	1
Dissolved Zinc	Filtered sample, ICP-MS, trace level. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	1
Bromide	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.05 g/m³	1
Chloride	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Nitrite-N	Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	1
Nitrate-N + Nitrite-N	Total oxidised nitrogen. Automated cadmium reduction, flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.002 g/m³	1
Sulphate	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	1
Ethylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID.	4 g/m³	1
Methanol in Water - Aqueous Solvents*	GC-FID analysis. In-house.	1.0 g/m³	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis. In-house based on US EPA 8260 and 5021.	0.0010 - 0.002 g/m³	1
Formaldehyde in Water by DNPH & LCMSMS	Derivatisation, SPE extraction, LC-MS/MS analysis. In-house based on US EPA 8315A.	0.02 g/m³	1
Gases in groundwater	Headspace GC-FID analysis. In-house.	0.002 - 0.003 g/m³	1
Total Petroleum Hydrocarbons in Water			
C7 - C9	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.10 g/m³	1
C10 - C14	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.2 g/m³	1
C15 - C36	Solvent extraction, GC-FID analysis. In-house based on US EPA 8015.	0.4 g/m³	1
Total hydrocarbons (C7 - C36)	Calculation: Sum of carbon bands from C7 to C36. In-house based on US EPA 8015.	0.7 g/m³	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Testing was completed between 11-Nov-2020 and 19-Nov-2020. For completion dates of individual analyses please contact the laboratory.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

Appendix IV

Certificates of analysis
(hydraulic fracturing fluids)

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

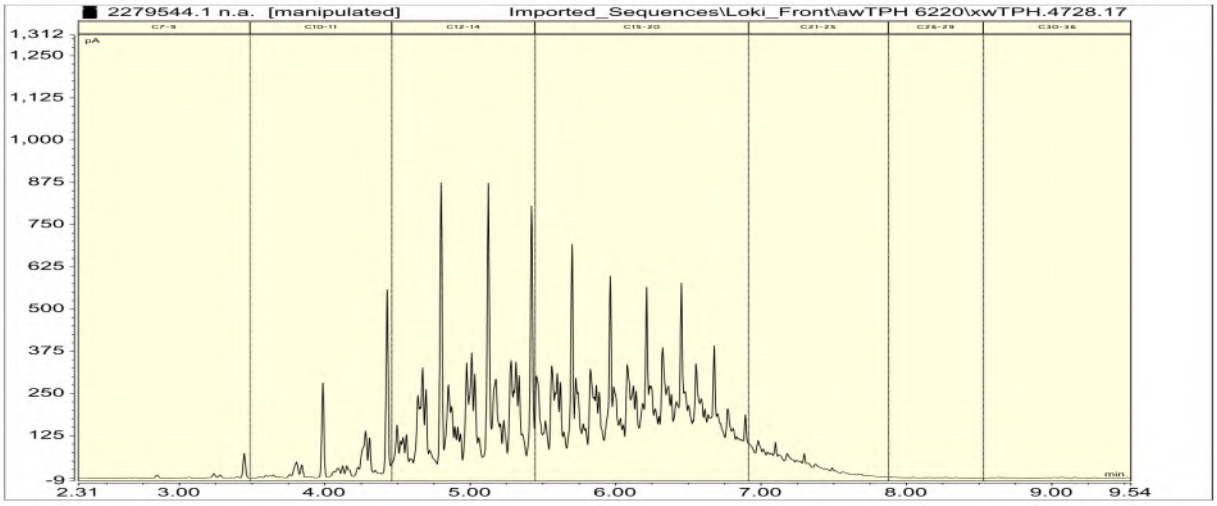
Certificate of Analysis

Client:	Greymouth Petroleum Limited	**Lab No:**	2279544 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	22-Nov-2019
	C/- Greymouth Petroleum Limited	**Date Reported:**	28-Nov-2019
	14 Connett Road West	**Quote No:**	85159
	Bell Block	**Order No:**	248338
	New Plymouth 4312	**Client Reference:**	Hydraulic fracturing fluid testing
		Submitted By:	Nakeysha Lammers

Sample Type: Aqueous

Sample Name:		Turangi 8 Stim 1 Pre-Pumped 14-Nov-2019				
Lab Number:		2279544.1				
Ethylene Glycol in Water						
Ethylene glycol*	g/m³	< 20	-	-	-	-
Propylene Glycol in Water						
Propylene glycol*	g/m³	< 20	-	-	-	-
Methanol in Water - Aqueous Solvents						
Methanol*	g/m³	< 2	-	-	-	-
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.010	-	-	-	-
Toluene	g/m³	< 0.010	-	-	-	-
Ethylbenzene	g/m³	< 0.010	-	-	-	-
m&p-Xylene	g/m³	0.02	-	-	-	-
o-Xylene	g/m³	0.012	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	8.2	-	-	-	-
C10 - C14	g/m³	690	-	-	-	-
C15 - C36	g/m³	1,210	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	1,910	-	-	-	-

2279544.1
Turangi 8 Stim 1 Pre-Pumped 14-Nov-2019
Client Chromatogram for TPH by FID



Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively clean matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous

Test	Method Description	Default Detection Limit	Sample No
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m^3	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m^3	1
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m^3	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Samples are held at the laboratory after reporting for a length of time depending on the preservation used and the stability of the analytes being tested. Once the storage period is completed the samples are discarded unless otherwise advised by the client.

Graham Corban MSc Tech (Hons)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

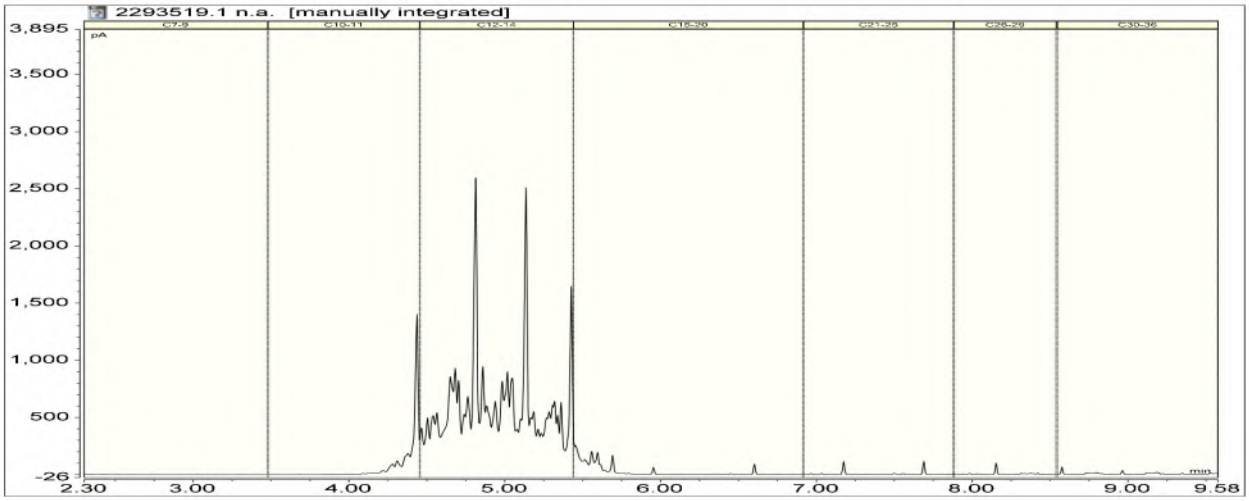
Certificate of Analysis

Client:	Greymouth Petroleum Limited	**Lab No:**	2293519 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	14-Dec-2019
	C/- Greymouth Petroleum Limited	**Date Reported:**	06-Jan-2020
	14 Connett Road West	**Quote No:**	85159
	Bell Block	**Order No:**	248639
	New Plymouth 4312	**Client Reference:**	Hydraulic fracturing fluid testing
		Submitted By:	Nakeysha Lammers

Sample Type: Aqueous

Sample Name:		Turangi 8 Stim 2 Pre-pumped 06-Dec-2019				
Lab Number:		2293519.1				
Ethylene Glycol in Water						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents						
Methanol*	g/m³	< 2	-	-	-	-
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.010	-	-	-	-
Toluene	g/m³	< 0.010	-	-	-	-
Ethylbenzene	g/m³	< 0.010	-	-	-	-
m&p-Xylene	g/m³	< 0.02	-	-	-	-
o-Xylene	g/m³	< 0.010	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	0.9	-	-	-	-
C10 - C14	g/m³	2,200	-	-	-	-
C15 - C36	g/m³	166	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	2,300	-	-	-	-

2293519.1
Turangi 8 Stim 2 Pre-pumped 06-Dec-2019
Client Chromatogram for TPH by FID



Summary of Methods

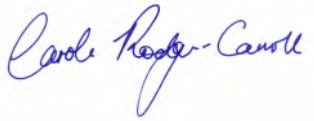
The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m^3	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m^3	1
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m^3	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.



Carole Rodgers-Carroll BA, NZCS
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

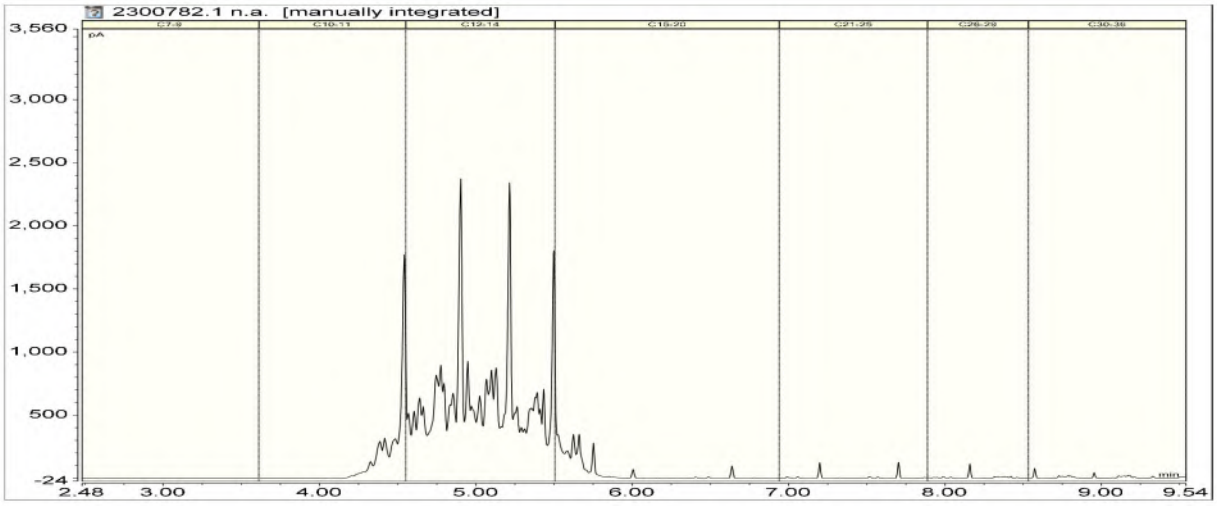
Certificate of Analysis

Client:	Greymouth Petroleum Limited	**Lab No:**	2300782 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	07-Jan-2020
	C/- Greymouth Petroleum Limited	**Date Reported:**	13-Jan-2020
	14 Connett Road West	**Quote No:**	85159
	Bell Block	**Order No:**	248714
	New Plymouth 4312	**Client Reference:**	Hydraulic fracturing fluid testing
		Submitted By:	Nakeysha Lammers

Sample Type: Aqueous

Sample Name:		Turangi 8 Stim 3 Pre-Pumped 14-Dec-2019				
Lab Number:		2300782.1				
Ethylene Glycol in Water						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents						
Methanol*	g/m³	2	-	-	-	-
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.010	-	-	-	-
Toluene	g/m³	0.023	-	-	-	-
Ethylbenzene	g/m³	< 0.010	-	-	-	-
m&p-Xylene	g/m³	< 0.02	-	-	-	-
o-Xylene	g/m³	< 0.010	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	< 0.8	-	-	-	-
C10 - C14	g/m³	2,700	-	-	-	-
C15 - C36	g/m³	290	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	3,000	-	-	-	-

2300782.1
Turangi 8 Stim 3 Pre-Pumped 14-Dec-2019
Client Chromatogram for TPH by FID



Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m^3	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m^3	1
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m^3	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Martin Cowell - BSc
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

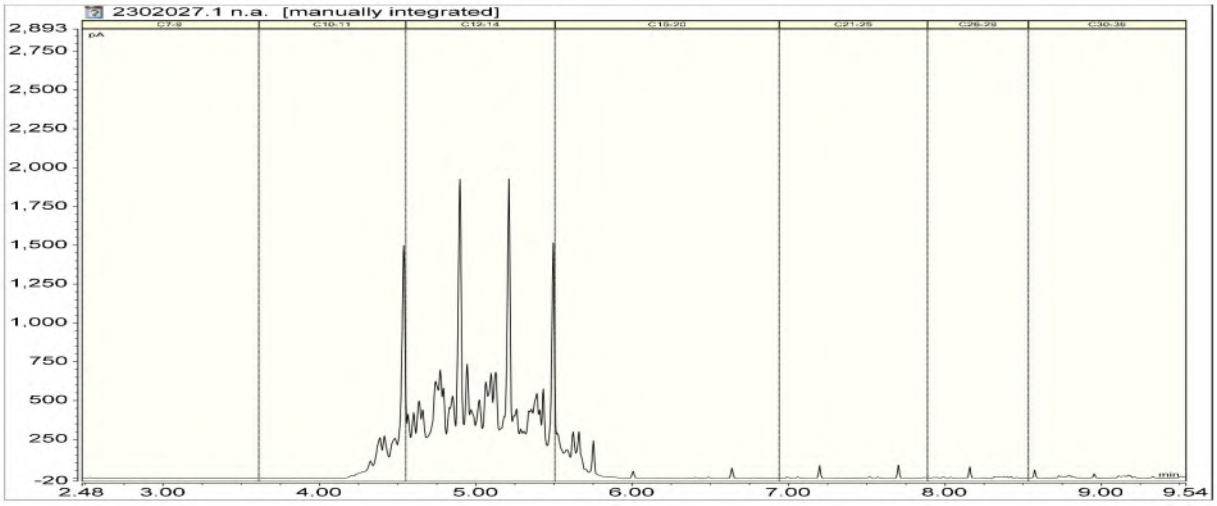
Certificate of Analysis

Client:	Greymouth Petroleum Limited	**Lab No:**	2302027 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	09-Jan-2020
	C/- Greymouth Petroleum Limited	**Date Reported:**	17-Jan-2020
	14 Connett Road West	**Quote No:**	85159
	Bell Block	**Order No:**	248845
	New Plymouth 4312	**Client Reference:**	Hydraulic fracturing fluid testing
		Submitted By:	Nakeysha Lammers

Sample Type: Aqueous

Sample Name:		Turangi 8 Stim 4 Pre-pumped 20-Dec-2019				
Lab Number:		2302027.1				
Ethylene Glycol in Water						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents						
Methanol*	g/m³	< 2	-	-	-	-
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	0.014	-	-	-	-
Toluene	g/m³	0.024	-	-	-	-
Ethylbenzene	g/m³	< 0.010	-	-	-	-
m&p-Xylene	g/m³	< 0.02	-	-	-	-
o-Xylene	g/m³	< 0.010	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	1.2	-	-	-	-
C10 - C14	g/m³	2,200	-	-	-	-
C15 - C36	g/m³	250	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	2,500	-	-	-	-

2302027.1
Turangi 8 Stim 4 Pre-pumped 20-Dec-2019
Client Chromatogram for TPH by FID



Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m^3	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m^3	1
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m^3	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Graham Corban MSc Tech (Hons)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

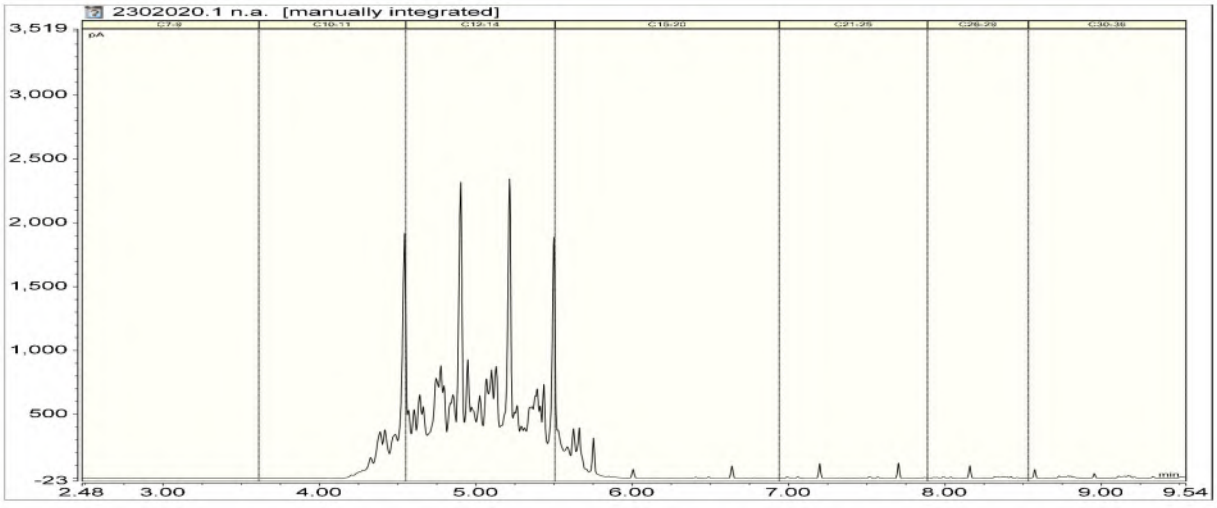
Certificate of Analysis

Client:	Greymouth Petroleum Limited	**Lab No:**	2302020 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	09-Jan-2020
	C/- Greymouth Petroleum Limited	**Date Reported:**	17-Jan-2020
	14 Connett Road West	**Quote No:**	85159
	Bell Block	**Order No:**	248844
	New Plymouth 4312	**Client Reference:**	Hydraulic fracturing fluid testing
		Submitted By:	Nakeysha Lammers

Sample Type: Aqueous

Sample Name:		Turangi 8 Stim 5 Pre-Pumped 27-Dec-2019				
Lab Number:		2302020.1				
Ethylene Glycol in Water						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents						
Methanol*	g/m³	< 2	-	-	-	-
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	0.012	-	-	-	-
Toluene	g/m³	0.014	-	-	-	-
Ethylbenzene	g/m³	< 0.010	-	-	-	-
m&p-Xylene	g/m³	< 0.02	-	-	-	-
o-Xylene	g/m³	< 0.010	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	1.0	-	-	-	-
C10 - C14	g/m³	2,900	-	-	-	-
C15 - C36	g/m³	330	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	3,200	-	-	-	-

2302020.1
Turangi 8 Stim 5 Pre-Pumped 27-Dec-2019
Client Chromatogram for TPH by FID



Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m^3	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m^3	1
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m^3	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Graham Corban MSc Tech (Hons)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

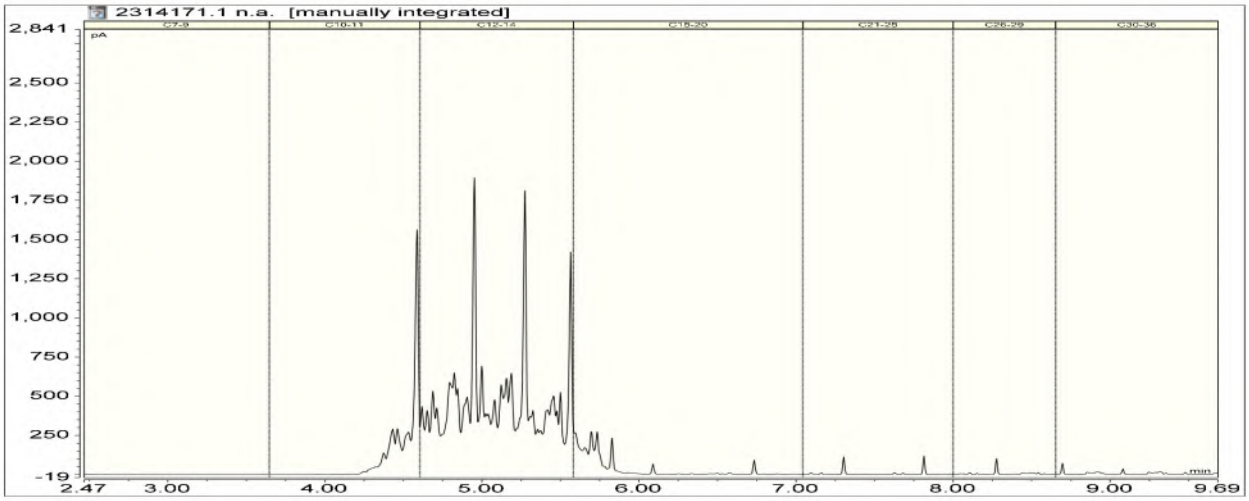
Certificate of Analysis

Client:	Greymouth Petroleum Limited	**Lab No:**	2314171 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	31-Jan-2020
	C/- Greymouth Petroleum Limited	**Date Reported:**	12-Feb-2020
	14 Connett Road West	**Quote No:**	85159
	Bell Block	**Order No:**	249098
	New Plymouth 4312	**Client Reference:**	Hydraulic fracturing fluid testing
		Submitted By:	Nakeysha Lammers

Sample Type: Aqueous

Sample Name:		Turangi 8 - Stim6 Prepumped ITF Fluid 25-Jan-2020				
Lab Number:		2314171.1				
Ethylene Glycol in Water*						
Ethylene glycol*	g/m³	< 4	-	-	-	-
Propylene Glycol in Water*						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*						
Methanol*	g/m³	< 2	-	-	-	-
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	< 0.010	-	-	-	-
Toluene	g/m³	< 0.010	-	-	-	-
Ethylbenzene	g/m³	< 0.010	-	-	-	-
m&p-Xylene	g/m³	< 0.02	-	-	-	-
o-Xylene	g/m³	< 0.010	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	0.6	-	-	-	-
C10 - C14	g/m³	1,930	-	-	-	-
C15 - C36	g/m³	240	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	2,200	-	-	-	-

2314171.1
Turangi 8 - Stim6 Prepumped ITF Fluid 25-Jan-2020
Client Chromatogram for TPH by FID



Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m^3	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m^3	1
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m^3	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

This certificate of analysis must not be reproduced, except in full, without the written consent of the signatory.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

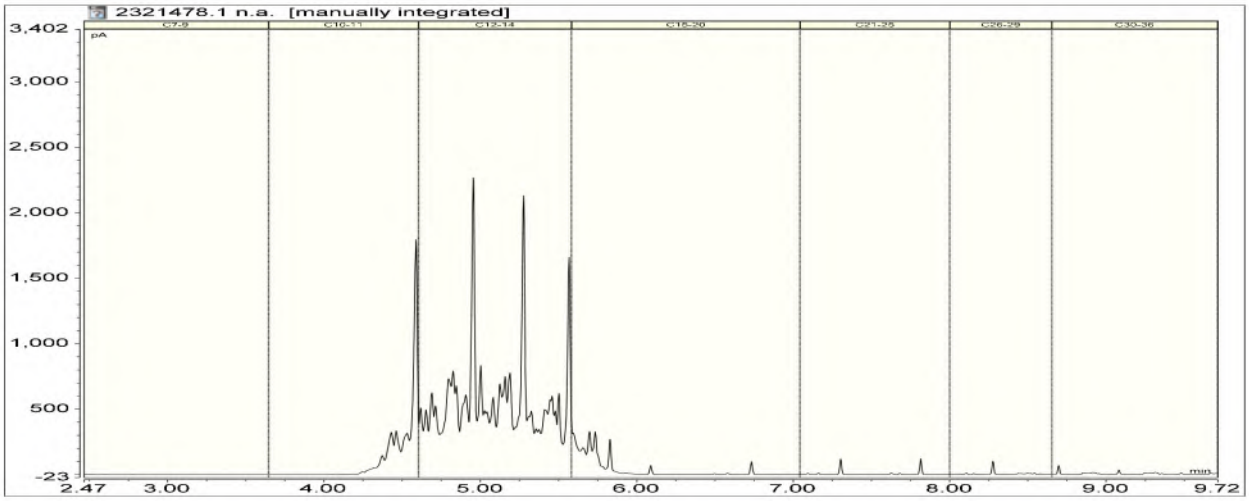
Certificate of Analysis

Client:	Greymouth Petroleum Limited	**Lab No:**	2321478 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	13-Feb-2020
	C/- Greymouth Petroleum Limited	**Date Reported:**	24-Feb-2020
	14 Connett Road West	**Quote No:**	85159
	Bell Block	**Order No:**	249267
	New Plymouth 4312	**Client Reference:**	Hydraulic fracturing fluid testing
		Submitted By:	Nakeysha Lammers

Sample Type: Aqueous

Sample Name:		Turangi 8 Stim 7 Prepumped 10-Feb-2020				
Lab Number:		2321478.1				
Ethylene Glycol in Water*						
Ethylene glycol*	g/m³	12	-	-	-	-
Propylene Glycol in Water*						
Propylene glycol*	g/m³	< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*						
Methanol*	g/m³	< 2	-	-	-	-
BTEX in Water by Headspace GC-MS						
Benzene	g/m³	0.013	-	-	-	-
Toluene	g/m³	0.029	-	-	-	-
Ethylbenzene	g/m³	< 0.010	-	-	-	-
m&p-Xylene	g/m³	< 0.02	-	-	-	-
o-Xylene	g/m³	< 0.010	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9	g/m³	0.4	-	-	-	-
C10 - C14	g/m³	2,300	-	-	-	-
C15 - C36	g/m³	250	-	-	-	-
Total hydrocarbons (C7 - C36)	g/m³	2,600	-	-	-	-

2321478.1
Turangi 8 Stim 7 Prepumped 10-Feb-2020
Client Chromatogram for TPH by FID



Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Aqueous			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m^3	1
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m^3	1
BTEX in Water by Headspace GC-MS	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m^3	1
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m^3	1

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

This certificate of analysis must not be reproduced, except in full, without the written consent of the signatory.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

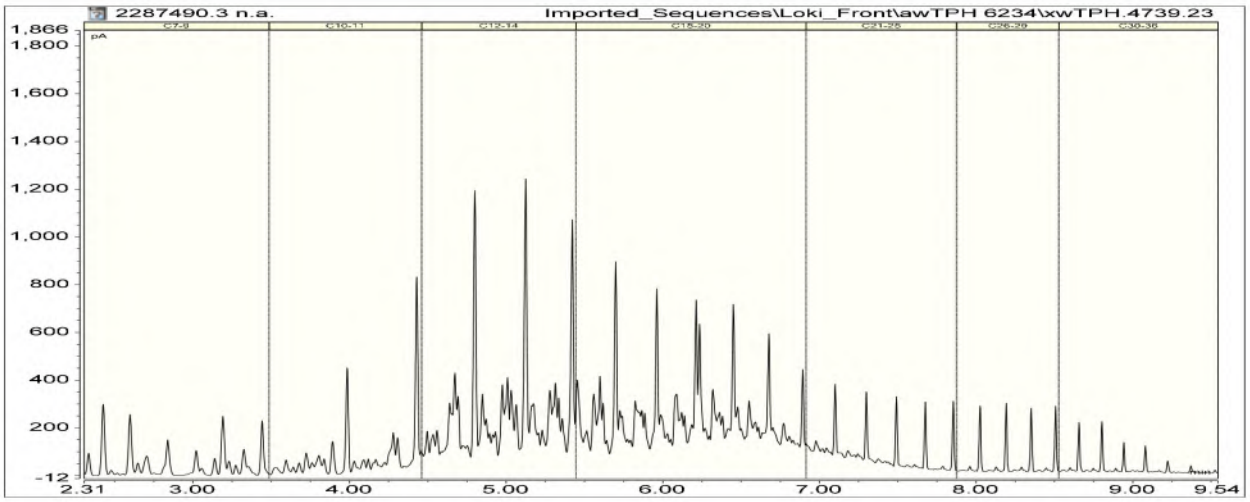
Client:	Greymouth Petroleum Limited	**Lab No:**	2287490 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	05-Dec-2019
	C/- Greymouth Petroleum Limited	**Date Reported:**	24-Dec-2019
	14 Connett Road West	**Quote No:**	81870
	Bell Block	**Order No:**	248338
	New Plymouth 4312	**Client Reference:**	Return Fluid Composite
		Submitted By:	Nakeysha Lammers

Sample Type: Saline

		Sample Name:	Composite of Turangi 8 - Stim 1 Composite Return Fluid				
		Lab Number:	2287490.3				
Individual Tests							
pH*	pH Units		6.9	-	-	-	-
Total Alkalinity*	g/m³ as $CaCO_3$		1,170	-	-	-	-
Total Hardness*	g/m³ as $CaCO_3$		88	-	-	-	-
Electrical Conductivity (EC)*	mS/m		474	-	-	-	-
Total Dissolved Solids (TDS)*	g/m³		4,300	-	-	-	-
Dissolved Barium	g/m³		1.29	-	-	-	-
Dissolved Bromine	g/m³		2.3	-	-	-	-
Dissolved Calcium	g/m³		25	-	-	-	-
Dissolved Copper	g/m³		0.022	-	-	-	-
Dissolved Iron	g/m³		0.68	-	-	-	-
Dissolved Magnesium	g/m³		6	-	-	-	-
Dissolved Manganese	g/m³		0.57	-	-	-	-
Total Nickel	g/m³		< 0.032	-	-	-	-
Total Potassium	g/m³		50	-	-	-	-
Total Sodium	g/m³		930	-	-	-	-
Total Sulphur*	g/m³		37	-	-	-	-
Total Zinc	g/m³		0.045	-	-	-	-
Chloride*	g/m³		790	-	-	-	-
Nitrite-N	g/m³		< 0.10 #2	-	-	-	-
Nitrate-N	g/m³		< 0.10	-	-	-	-
Nitrate*	g/m³		< 0.5	-	-	-	-
Nitrate-N + Nitrite-N	g/m³		< 0.10 #2	-	-	-	-
Sulphate*	g/m³		111	-	-	-	-
Ethylene Glycol in Water							
Ethylene glycol*	g/m³		< 20	-	-	-	-
Propylene Glycol in Water							
Propylene glycol*	g/m³		< 20	-	-	-	-
Methanol in Water - Aqueous Solvents							
Methanol*	g/m³		< 5 #1	-	-	-	-
BTEX in Water by Headspace GC-MS							
Benzene*	g/m³		7.9	-	-	-	-
Toluene*	g/m³		9.8	-	-	-	-
Ethylbenzene*	g/m³		1.22	-	-	-	-
m&p-Xylene*	g/m³		4.7	-	-	-	-
o-Xylene*	g/m³		1.99	-	-	-	-

Sample Type: Saline						
Sample Name:		Composite of Turangi 8 - Stim 1 Composite Return Fluid				
Lab Number:		2287490.3				
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde*	g/m³	< 0.15	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9*	g/m³	64	-	-	-	-
C10 - C14*	g/m³	410	-	-	-	-
C15 - C36*	g/m³	660	-	-	-	-
Total hydrocarbons (C7 - C36)*	g/m³	1,130	-	-	-	-

2287490.3
Composite of Turangi 8 - Stim 1 Composite Return Fluid
Client Chromatogram for TPH by FID



Analyst's Comments

#1 Due to some interference found in the chromatography, the detection limit was raised. Hence the higher detection limit reported.

#2 Due to the nature of this sample a dilution was performed prior to analysis, resulting in a detection limit higher than that normally achieved for the NO2N, NO3N and NOxN analysis.

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	3
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	3
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m³	3
BTEX in Water by Headspace GC-MS*	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m³	3
Formaldehyde in Water by DNPH & LCMSMS*	DNPH derivatisation, extraction, LCMSMS	0.02 g/m³	3
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m³	3
Filtration, Unpreserved*	Sample filtration through 0.45µm membrane filter.	-	3
Total Digestion*	Boiling nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	3
Total Digestion of Saline Samples*	Nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	3

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
pH*	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	3
Total Alkalinity*	Saline water, Titration to pH 4.5.	1.0 g/m^3 as $CaCO_3$	3
Total Hardness*	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m^3 as $CaCO_3$	3
Electrical Conductivity (EC)*	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	3
Total Dissolved Solids (TDS)*	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	50 g/m^3	3
Filtration for dissolved metals analysis*	Sample filtration through 0.45µm membrane filter and preservation with nitric acid. APHA 3030 B 23rd ed. 2017.	-	3
Dissolved Barium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0006 g/m^3	3
Dissolved Bromine	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.10 g/m^3	3
Dissolved Calcium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.0 g/m^3	3
Dissolved Copper	Filtered sample, ICP-MS, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m^3	3
Dissolved Iron	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.004 g/m^3	3
Dissolved Magnesium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.4 g/m^3	3
Dissolved Manganese	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m^3	3
Total Nickel	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0070 g/m^3	3
Total Potassium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.1 g/m^3	3
Total Sodium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.42 g/m^3	3
Total Sulphur*	Nitric acid digestion, ICP-OES (method may not fully account for H_2S due to volatilisation during digestion). All forms of oxidised and organic sulphur will be determined by this method. APHA 3120 B 23rd ed. 2017.	0.5 g/m^3	3
Total Zinc	Nitric acid digestion, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0042 g/m^3	3
Chloride*	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m^3	3
Nitrite-N	Saline sample. Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m^3	3
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m^3	3
Nitrate*	Calculation from Nitrate-N.	0.005 g/m^3	3
Nitrate-N + Nitrite-N	Saline sample. Total oxidised nitrogen. Automated cadmium reduction, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m^3	3
Total Sulphate*	Calculation: from total sulphur.	2 g/m^3	3

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Helena Bertram BSc
Client Services Manager - Environmental

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

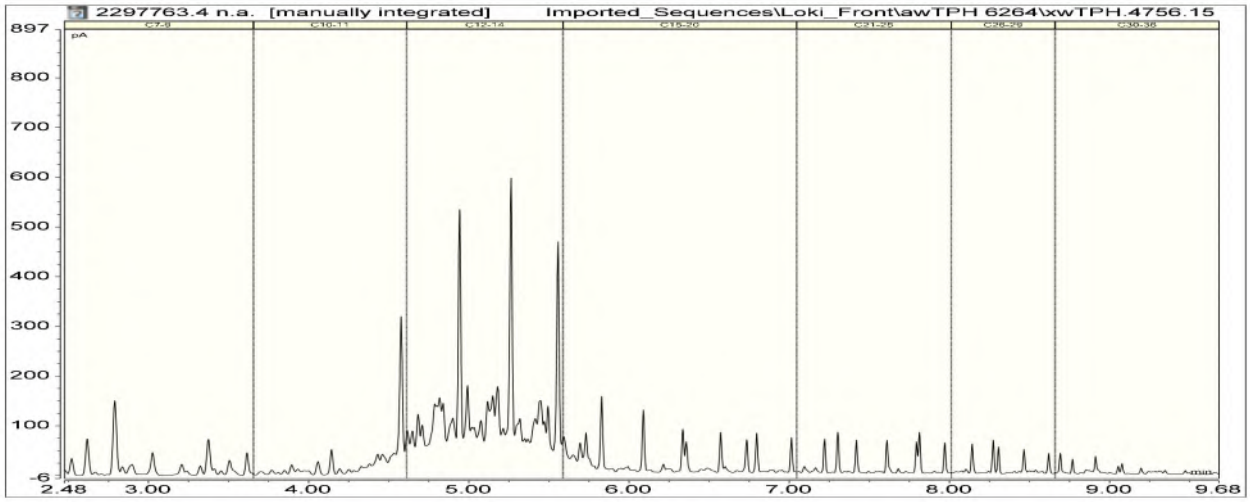
Client:	Greymouth Petroleum Limited	**Lab No:**	2297763 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	21-Dec-2019
	C/- Greymouth Petroleum Limited	**Date Reported:**	16-Jan-2020
	14 Connett Road West	**Quote No:**	81870
	Bell Block	**Order No:**	248639
	New Plymouth 4312	**Client Reference:**	Return Fluid Composite
		Submitted By:	Nakeysha Lammers

Sample Type: Saline

Sample Name:		Turangi 8 Stim 2 Composite Return Fluid				
Lab Number:		2297763.4				
Individual Tests						
pH*	pH Units	6.3	-	-	-	-
Total Alkalinity*	g/m³ as $CaCO_3$	820	-	-	-	-
Analysis Temperature for Bicarbonate	°C	22	-	-	-	-
Bicarbonate	g/m³ at Analysis Temperature	684	-	-	-	-
Total Hardness*	g/m³ as $CaCO_3$	143	-	-	-	-
Electrical Conductivity (EC)*	mS/m	1,301	-	-	-	-
Total Dissolved Solids (TDS)*	g/m³	9,000	-	-	-	-
Dissolved Barium	g/m³	28	-	-	-	-
Dissolved Bromine	g/m³	11.6	-	-	-	-
Dissolved Calcium	g/m³	48	-	-	-	-
Dissolved Copper	g/m³	< 0.005	-	-	-	-
Dissolved Iron	g/m³	2.6	-	-	-	-
Dissolved Magnesium	g/m³	6	-	-	-	-
Dissolved Manganese	g/m³	2.4	-	-	-	-
Total Nickel	g/m³	0.035	-	-	-	-
Total Potassium	g/m³	184	-	-	-	-
Total Sodium	g/m³	2,800	-	-	-	-
Total Sulphur*	g/m³	21	-	-	-	-
Total Zinc	g/m³	0.037	-	-	-	-
Chloride*	g/m³	3,700	-	-	-	-
Nitrite-N	g/m³	< 0.10 #1	-	-	-	-
Nitrate-N	g/m³	< 0.10	-	-	-	-
Nitrate*	g/m³	< 0.5	-	-	-	-
Nitrate-N + Nitrite-N	g/m³	< 0.10 #1	-	-	-	-
Sulphate*	g/m³	64	-	-	-	-
Ethylene Glycol in Water						
Ethylene glycol*	g/m³	< 20	-	-	-	-
Propylene Glycol in Water						
Propylene glycol*	g/m³	< 20	-	-	-	-
Methanol in Water - Aqueous Solvents						
Methanol*	g/m³	< 20	-	-	-	-
BTEX in Water by Headspace GC-MS						
Benzene*	g/m³	11.9	-	-	-	-
Toluene*	g/m³	7.8	-	-	-	-
Ethylbenzene*	g/m³	0.54	-	-	-	-
m&p-Xylene*	g/m³	2.2	-	-	-	-
o-Xylene*	g/m³	0.89	-	-	-	-

Sample Type: Saline						
Sample Name:		Turangi 8 Stim 2 Composite Return Fluid				
Lab Number:		2297763.4				
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde*	g/m³	< 0.15	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9*	g/m³	21	-	-	-	-
C10 - C14*	g/m³	174	-	-	-	-
C15 - C36*	g/m³	93	-	-	-	-
Total hydrocarbons (C7 - C36)*	g/m³	290	-	-	-	-

2297763.4
Turangi 8 Stim 2 Composite Return Fluid
Client Chromatogram for TPH by FID



Analyst's Comments

#1 Due to the nature of this sample that a dilution be performed prior to analysis, resulting in a detection limit higher than that normally achieved for the NO2N, NO3N and NOxN analysis.

Appendix No.1 - GNS report

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	4
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	4
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m³	4
BTEX in Water by Headspace GC-MS*	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m³	4
Formaldehyde in Water by DNPH & LCMSMS*	DNPH derivatisation, extraction, LCMSMS	0.02 g/m³	4
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m³	4
Filtration, Unpreserved*	Sample filtration through 0.45µm membrane filter.	-	4
Total Digestion*	Boiling nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	4
Total Digestion of Saline Samples*	Nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	4

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
pH*	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	4
Total Alkalinity*	Saline water, Titration to pH 4.5.	1.0 g/m³ as $CaCO_3$	4
Analysis Temperature for Bicarbonate	Temperature at which Bicarbonate titration was conducted as reported by Geological & Nuclear Sciences, Wairakei.	1.0 °C	4
Bicarbonate	Bicarbonate (HCO3) Titration Method conducted at reported temperature. Subcontracted to Geological & Nuclear Sciences, Wairakei. ASTM Standards D513-82 Vol.11.01 of 1988.	20 g/m³ at Analysis Temperature	4
Total Hardness*	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	4
Electrical Conductivity (EC)*	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	4
Total Dissolved Solids (TDS)*	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	50 g/m³	4
Filtration for dissolved metals analysis*	Sample filtration through 0.45µm membrane filter and preservation with nitric acid. APHA 3030 B 23rd ed. 2017.	-	4
Dissolved Barium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0006 g/m³	4
Dissolved Bromine	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.10 g/m³	4
Dissolved Calcium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.0 g/m³	4
Dissolved Copper	Filtered sample, ICP-MS, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	4
Dissolved Iron	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.004 g/m³	4
Dissolved Magnesium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.4 g/m³	4
Dissolved Manganese	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	4
Total Nickel	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0070 g/m³	4
Total Potassium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.1 g/m³	4
Total Sodium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.42 g/m³	4
Total Sulphur*	Nitric acid digestion, ICP-OES (method may not fully account for H_2S due to volatilisation during digestion). All forms of oxidised and organic sulphur will be determined by this method. APHA 3120 B 23rd ed. 2017.	0.5 g/m³	4
Total Zinc	Nitric acid digestion, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0042 g/m³	4
Chloride*	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	4
Nitrite-N	Saline sample. Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m³	4
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	4
Nitrate*	Calculation from Nitrate-N.	0.005 g/m³	4
Nitrate-N + Nitrite-N	Saline sample. Total oxidised nitrogen. Automated cadmium reduction, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m³	4
Total Sulphate*	Calculation: from total sulphur.	2 g/m³	4

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

GNS SCIENCE TE PŪ AO
ANALYTICAL LABORATORY
Private Bag 2000, Taupo
Phone: (07) 374 8211
Fax: (07) 374 8199
Email: w.labmanager@gns.cri.nz

CERTIFICATE OF ANALYSIS

ENVSUBGNS_WAIRAKEI 104

Report No: 2020010709

Customer Ref:153091

Ara Heron
RJ Hill Laboratories (Hamilton)
Environmental Reports Officers
Private Bag 3205
Hamilton

GNS Lot No: 2020010709

GNS Sample No.	2020000072
Collection Date	
Site ID	2297763-4
Field ID	

pH		6.49	-	-	-
Bicarbonate (Total)	mg/l	684	-	-	-
HCO_3 Analysis Temperature	°C	22	-	-	-
HCO_3 Analysis Date		07/01/2020	-	-	-

SUMMARY OF METHODS AND DETECTION LIMITS

The following table gives a brief description of the methods used to conduct the analyses on this report.
The detection limits given below are those attainable in a relatively clean matrix.

Parameter	Method	*Detection Limit	
Bicarbonate (total)	HCO_3 Titration Method ASTM Standards D513-82 Vol.11.01 1988	20	mg/l
pH	Electrometric Method - APHA 4500-H+ B 23rd Edition 2017	-	-

*Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis.

Notes: These samples were collected by yourselves (or your agent) and analysed as received at the laboratory. This report must not be reproduced, except in full, without the written consent of the signatory. Samples are held at the laboratory after reporting for a period of 2 to 6 months, dependent on sample type.

M. K. Appleby

Moya Appleby
Principal Technician

ilac-MRA IANZ ACCREDITED LABORATORY Tests marked with a † are not accredited and are outside the scope of the laboratory's accreditation

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

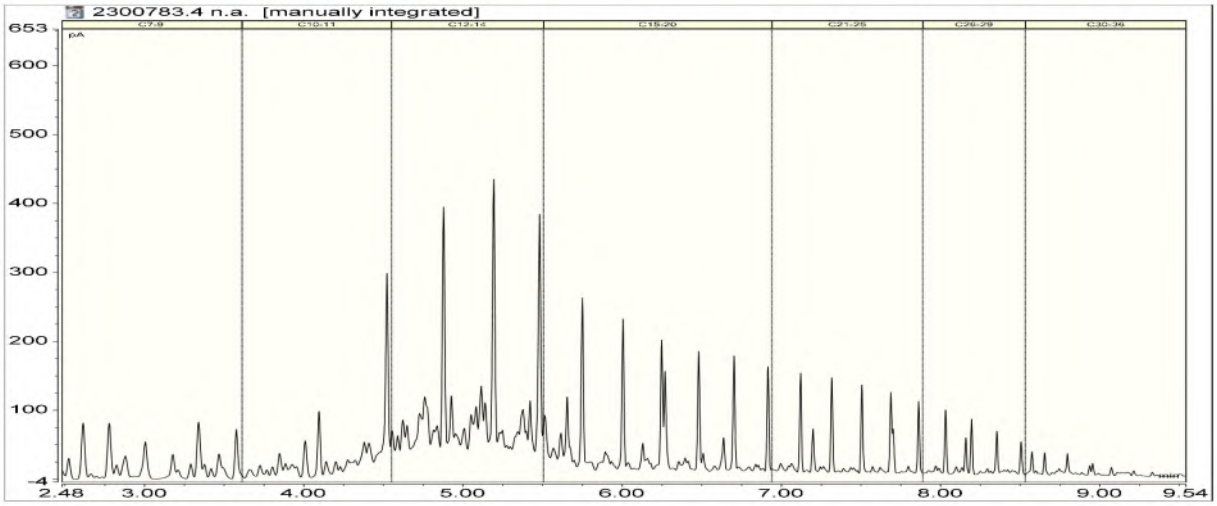
Client:	Greymouth Petroleum Limited	**Lab No:**	2300783 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	07-Jan-2020
	C/- Greymouth Petroleum Limited	**Date Reported:**	22-Jan-2020
	14 Connett Road West	**Quote No:**	81870
	Bell Block	**Order No:**	248714
	New Plymouth 4312	**Client Reference:**	Return Fluid Composite
		Submitted By:	Nakeysha Lammers

Sample Type: Saline

Sample Name:		Turangi-8 Stim:3 Composite Return Fluid				
Lab Number:		2300783.4				
Individual Tests						
pH*	pH Units	6.4	-	-	-	-
Total Alkalinity*	g/m³ as $CaCO_3$	610	-	-	-	-
Analysis Temperature for Bicarbonate	°C	22	-	-	-	-
Bicarbonate	g/m³ at Analysis Temperature	439	-	-	-	-
Total Hardness*	g/m³ as $CaCO_3$	400	-	-	-	-
Electrical Conductivity (EC)*	mS/m	1,824	-	-	-	-
Total Dissolved Solids (TDS)*	g/m³	12,100	-	-	-	-
Dissolved Barium	g/m³	37	-	-	-	-
Dissolved Bromine	g/m³	16.5	-	-	-	-
Dissolved Calcium	g/m³	126	-	-	-	-
Dissolved Copper	g/m³	0.017	-	-	-	-
Dissolved Iron	g/m³	3.9	-	-	-	-
Dissolved Magnesium	g/m³	20	-	-	-	-
Dissolved Manganese	g/m³	3.1	-	-	-	-
Total Nickel	g/m³	< 0.032	-	-	-	-
Total Potassium	g/m³	290	-	-	-	-
Total Sodium	g/m³	3,900	-	-	-	-
Total Sulphur*	g/m³	48	-	-	-	-
Total Zinc	g/m³	0.151	-	-	-	-
Chloride*	g/m³	6,200	-	-	-	-
Nitrite-N	g/m³	< 0.10 #1	-	-	-	-
Nitrate-N	g/m³	< 0.10	-	-	-	-
Nitrate*	g/m³	< 0.5	-	-	-	-
Nitrate-N + Nitrite-N	g/m³	< 0.10 #1	-	-	-	-
Sulphate*	g/m³	145	-	-	-	-
Ethylene Glycol in Water						
Ethylene glycol*	g/m³	43	-	-	-	-
Propylene Glycol in Water						
Propylene glycol*	g/m³	< 20	-	-	-	-
Methanol in Water - Aqueous Solvents						
Methanol*	g/m³	< 20	-	-	-	-
BTEX in Water by Headspace GC-MS						
Benzene*	g/m³	6.4	-	-	-	-
Toluene*	g/m³	6.2	-	-	-	-
Ethylbenzene*	g/m³	0.79	-	-	-	-
m&p-Xylene*	g/m³	3.6	-	-	-	-
o-Xylene*	g/m³	1.49	-	-	-	-

Sample Type: Saline						
Sample Name:		Turangi-8 Stim:3 Composite Return Fluid				
Lab Number:		2300783.4				
Formaldehyde in Water by DNPH & LCMSMS						
Formaldehyde*	g/m³	< 0.15	-	-	-	-
Total Petroleum Hydrocarbons in Water						
C7 - C9*	g/m³	29	-	-	-	-
C10 - C14*	g/m³	167	-	-	-	-
C15 - C36*	g/m³	161	-	-	-	-
Total hydrocarbons (C7 - C36)*	g/m³	360	-	-	-	-

2300783.4
Turangi-8 Stim:3 Composite Return Fluid
Client Chromatogram for TPH by FID



Analyst's Comments

#1 Due to the nature of this sample a dilution was performed prior to analysis, resulting in a detection limit higher than that normally achieved for the NO2N, NO3N and NOxN analys

Appendix No.1 - GNS Report

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	4
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	4
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m³	4
BTEX in Water by Headspace GC-MS*	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m³	4
Formaldehyde in Water by DNPH & LCMSMS*	DNPH derivatisation, extraction, LCMSMS	0.02 g/m³	4
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m³	4
Filtration, Unpreserved*	Sample filtration through 0.45µm membrane filter.	-	4
Total Digestion*	Boiling nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	4
Total Digestion of Saline Samples*	Nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	4

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
pH*	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	4
Total Alkalinity*	Saline water, Titration to pH 4.5.	1.0 g/m³ as $CaCO_3$	4
Analysis Temperature for Bicarbonate	Temperature at which Bicarbonate titration was conducted as reported by Geological & Nuclear Sciences, Wairakei.	1.0 °C	4
Bicarbonate	Bicarbonate (HCO3) Titration Method conducted at reported temperature. Subcontracted to Geological & Nuclear Sciences, Wairakei. ASTM Standards D513-82 Vol.11.01 of 1988.	20 g/m³ at Analysis Temperature	4
Total Hardness*	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	4
Electrical Conductivity (EC)*	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	4
Total Dissolved Solids (TDS)*	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	50 g/m³	4
Filtration for dissolved metals analysis*	Sample filtration through 0.45µm membrane filter and preservation with nitric acid. APHA 3030 B 23rd ed. 2017.	-	4
Dissolved Barium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0006 g/m³	4
Dissolved Bromine	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.10 g/m³	4
Dissolved Calcium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.0 g/m³	4
Dissolved Copper	Filtered sample, ICP-MS, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	4
Dissolved Iron	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.004 g/m³	4
Dissolved Magnesium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.4 g/m³	4
Dissolved Manganese	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	4
Total Nickel	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0070 g/m³	4
Total Potassium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.1 g/m³	4
Total Sodium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.42 g/m³	4
Total Sulphur*	Nitric acid digestion, ICP-OES (method may not fully account for H_2S due to volatilisation during digestion). All forms of oxidised and organic sulphur will be determined by this method. APHA 3120 B 23rd ed. 2017.	0.5 g/m³	4
Total Zinc	Nitric acid digestion, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0042 g/m³	4
Chloride*	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	4
Nitrite-N	Saline sample. Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m³	4
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	4
Nitrate*	Calculation from Nitrate-N.	0.005 g/m³	4
Nitrate-N + Nitrite-N	Saline sample. Total oxidised nitrogen. Automated cadmium reduction, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m³	4
Total Sulphate*	Calculation: from total sulphur.	2 g/m³	4

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Graham Corban MSc Tech (Hons)
Client Services Manager - Environmental

ANALYTICAL LABORATORY
Private Bag 2000, Taupo
Phone: (07) 374 8211
Fax: (07) 374 8199
Email: w.labmanager@gns.cri.nz

CERTIFICATE OF ANALYSIS
ENVSUBGNS_WAIRAKEI 105

Report No: 2020010902

Customer Ref:153111

Ara Heron
RJ Hill Laboratories (Hamilton)
Environmental Reports Officers
Private Bag 3205
Hamilton

GNS Lot No: 2020010902

GNS Sample No.	2020000123
Collection Date	
Site ID	2300783.4
Field ID	

pH		6.32	-	-	-
Bicarbonate (Total)	mg/l	439	-	-	-
HCO_3 Analysis Temperature	°C	22	-	-	-
HCO_3 Analysis Date		13/01/2020	-	-	-

SUMMARY OF METHODS AND DETECTION LIMITS

The following table gives a brief description of the methods used to conduct the analyses on this report.
The detection limits given below are those attainable in a relatively clean matrix.

Parameter	Method	*Detection Limit	
Bicarbonate (total)	HCO_3 Titration Method ASTM Standards D513-82 Vol.11.01 1988	20	mg/l
pH	Electrometric Method - APHA 4500-H+ B 23rd Edition 2017	-	-

*Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis.

Notes: These samples were collected by yourselves (or your agent) and analysed as received at the laboratory. This report must not be reproduced, except in full, without the written consent of the signatory. Samples are held at the laboratory after reporting for a period of 2 to 6 months, dependent on sample type.

M. K. Appleby

Moya Appleby
Principal Technician

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

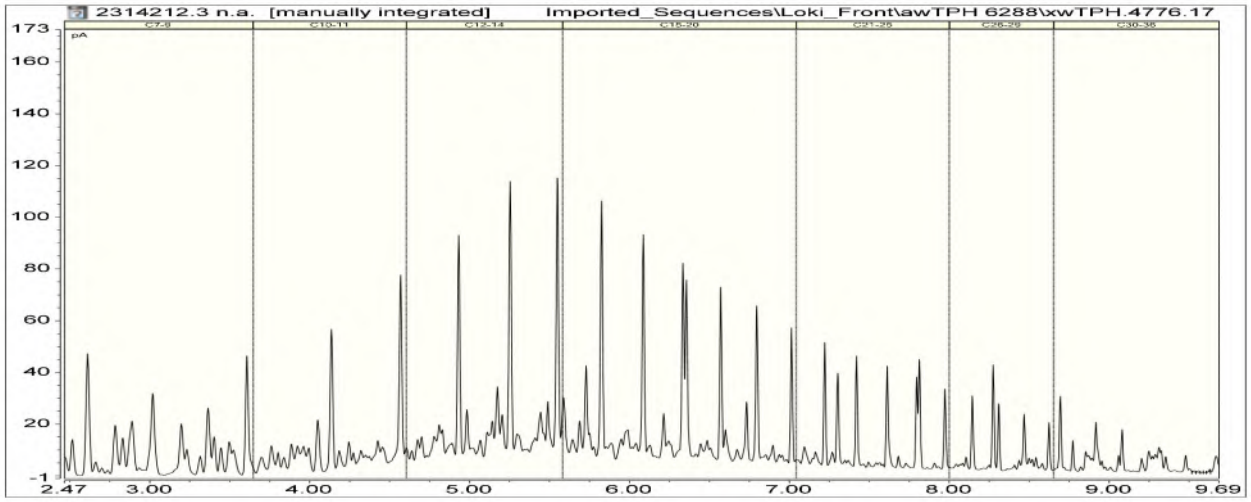
Client:	Greymouth Petroleum Limited	**Lab No:**	2314212 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	31-Jan-2020
	C/- Greymouth Petroleum Limited	**Date Reported:**	18-Feb-2020
	14 Connett Road West	**Quote No:**	81870
	Bell Block	**Order No:**	248845
	New Plymouth 4312	**Client Reference:**	Return Fluid Composite
		Submitted By:	Nakeysha Lammers

Sample Type: Saline

		Sample Name:	Turangi 8 Stim 4 Composite Return Fluid				
		Lab Number:	2314212.3				
Individual Tests							
pH*	pH Units		5.7	-	-	-	-
Total Alkalinity*	g/m³ as $CaCO_3$		280	-	-	-	-
Analysis Temperature for Bicarbonate‡	°C		21	-	-	-	-
Bicarbonate‡	g/m³ at Analysis Temperature		82	-	-	-	-
Total Hardness*	g/m³ as $CaCO_3$		189	-	-	-	-
Electrical Conductivity (EC)*	mS/m		797	-	-	-	-
Total Dissolved Solids (TDS)*	g/m³		6,200	-	-	-	-
Dissolved Barium	g/m³		8.1	-	-	-	-
Dissolved Bromine	g/m³		6.3	-	-	-	-
Dissolved Calcium	g/m³		58	-	-	-	-
Dissolved Copper	g/m³		0.076	-	-	-	-
Dissolved Iron	g/m³		1.18	-	-	-	-
Dissolved Magnesium	g/m³		11	-	-	-	-
Dissolved Manganese	g/m³		3.3	-	-	-	-
Total Nickel	g/m³		0.051	-	-	-	-
Total Potassium	g/m³		197	-	-	-	-
Total Sodium	g/m³		1,490	-	-	-	-
Total Sulphur*	g/m³		45	-	-	-	-
Total Zinc	g/m³		0.084 #1	-	-	-	-
Chloride*	g/m³		2,400	-	-	-	-
Nitrite-N	g/m³		< 0.10 #2	-	-	-	-
Nitrate-N	g/m³		< 0.10	-	-	-	-
Nitrate*	g/m³		< 0.5	-	-	-	-
Nitrate-N + Nitrite-N	g/m³		< 0.10 #2	-	-	-	-
Sulphate*	g/m³		133	-	-	-	-
Ethylene Glycol in Water*							
Ethylene glycol*	g/m³		16	-	-	-	-
Propylene Glycol in Water*							
Propylene glycol*	g/m³		< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*							
Methanol*	g/m³		< 2	-	-	-	-
BTEX in Water by Headspace GC-MS*							
Benzene*	g/m³		2.3	-	-	-	-
Toluene*	g/m³		2.6	-	-	-	-
Ethylbenzene*	g/m³		0.38	-	-	-	-
m&p-Xylene*	g/m³		2.2	-	-	-	-
o-Xylene*	g/m³		1.02	-	-	-	-

Sample Type: Saline						
Sample Name:	Turangi 8 Stim 4 Composite Return Fluid					
Lab Number:	2314212.3					
Formaldehyde in Water by DNPH & LCMSMS*						
Formaldehyde* g/m³	< 0.15	-	-	-	-	
Total Petroleum Hydrocarbons in Water*						
C7 - C9* g/m³	34	-	-	-	-	
C10 - C14* g/m³	95	-	-	-	-	
C15 - C36* g/m³	150	-	-	-	-	
Total hydrocarbons (C7 - C36)* g/m³	280	-	-	-	-	

2314212.3
Turangi 8 Stim 4 Composite Return Fluid
Client Chromatogram for TPH by FID



Analyst's Comments

‡ Analysis subcontracted to an external provider. Refer to the Summary of Methods section for more details.

#1 It should be noted that the replicate analyses performed on this sample as part of our in-house Quality Assurance procedures showed greater variation than would normally be expected. This may reflect the heterogeneity of the sample. The average of the results of the replicate analyses has been reported.

#2 Due to the nature of this sample a dilution was performed prior to analysis, resulting in a detection limit higher than that normally achieved for the NO2N, NO3N and NOxN analysis.

Appendix No.1 - GNS report

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	3
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	3
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m³	3
BTEX in Water by Headspace GC-MS*	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m³	3
Formaldehyde in Water by DNPH & LCMSMS*	DNPH derivatisation, extraction, LCMSMS	0.02 g/m³	3
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m³	3
Filtration, Unpreserved*	Sample filtration through 0.45µm membrane filter.	-	3
Total Digestion*	Boiling nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	3

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Total Digestion of Saline Samples*	Nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	3
pH*	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	3
Total Alkalinity*	Saline water, Titration to pH 4.5.	1.0 g/m^3 as $CaCO_3$	3
Analysis Temperature for Bicarbonate	Temperature at which Bicarbonate titration was conducted as reported by Geological & Nuclear Sciences, Wairakei.	1.0 °C	3
Bicarbonate	Bicarbonate (HCO3) Titration Method conducted at reported temperature. Subcontracted to Geological & Nuclear Sciences, Wairakei. ASTM Standards D513-82 Vol.11.01 of 1988.	20 g/m^3 at Analysis Temperature	3
Total Hardness*	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m^3 as $CaCO_3$	3
Electrical Conductivity (EC)*	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	3
Total Dissolved Solids (TDS)*	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	50 g/m^3	3
Filtration for dissolved metals analysis*	Sample filtration through 0.45µm membrane filter and preservation with nitric acid. APHA 3030 B 23rd ed. 2017.	-	3
Dissolved Barium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0006 g/m^3	3
Dissolved Bromine	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.10 g/m^3	3
Dissolved Calcium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.0 g/m^3	3
Dissolved Copper	Filtered sample, ICP-MS, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m^3	3
Dissolved Iron	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.004 g/m^3	3
Dissolved Magnesium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.4 g/m^3	3
Dissolved Manganese	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m^3	3
Total Nickel	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0070 g/m^3	3
Total Potassium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.1 g/m^3	3
Total Sodium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.42 g/m^3	3
Total Sulphur*	Nitric acid digestion, ICP-OES (method may not fully account for H_2S due to volatilisation during digestion). All forms of oxidised and organic sulphur will be determined by this method. APHA 3120 B 23rd ed. 2017.	0.5 g/m^3	3
Total Zinc	Nitric acid digestion, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0042 g/m^3	3
Chloride*	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m^3	3
Nitrite-N	Saline sample. Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m^3	3
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m^3	3
Nitrate*	Calculation from Nitrate-N.	0.005 g/m^3	3
Nitrate-N + Nitrite-N	Saline sample. Total oxidised nitrogen. Automated cadmium reduction, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m^3	3
Total Sulphate*	Calculation: from total sulphur.	2 g/m^3	3

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

GNS SCIENCE TE PŪ AO
ANALYTICAL LABORATORY
Private Bag 2000, Taupo
Phone: (07) 374 8211
Fax: (07) 374 8199
Email: w.labmanager@gns.cri.nz

CERTIFICATE OF ANALYSIS

ENVSUBGNS_WAIRAKEI 115

Report No: 2020020408

Customer Ref:153312

Ara Heron
RJ Hill Laboratories (Hamilton)
Environmental Reports Officers
Private Bag 3205
Hamilton

GNS Lot No: 2020020408

GNS Sample No.	2020000670
Collection Date	
Site ID	2314212.3
Field ID	

pH		5.81	-	-	-
Bicarbonate (Total)	mg/l	82	-	-	-
HCO_3 Analysis Temperature	°C	21	-	-	-
HCO_3 Analysis Date		05/02/2020	-	-	-

SUMMARY OF METHODS AND DETECTION LIMITS

The following table gives a brief description of the methods used to conduct the analyses on this report.
The detection limits given below are those attainable in a relatively clean matrix.

Parameter	Method	*Detection Limit	
Bicarbonate (total)	HCO_3 Titration Method ASTM Standards D513-82 Vol.11.01 1988	20	mg/l
pH	Electrometric Method - APHA 4500-H+ B 23rd Edition 2017	-	-

*Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis.

Notes: These samples were collected by yourselves (or your agent) and analysed as received at the laboratory. This report must not be reproduced, except in full, without the written consent of the signatory. Samples are held at the laboratory after reporting for a period of 2 to 6 months, dependent on sample type.

A Noddings
Ann Noddings
Senior Technician

ilac-MRA IANZ ACCREDITED LABORATORY Tests marked with a † are not accredited and are outside the scope of the laboratory's accreditation

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

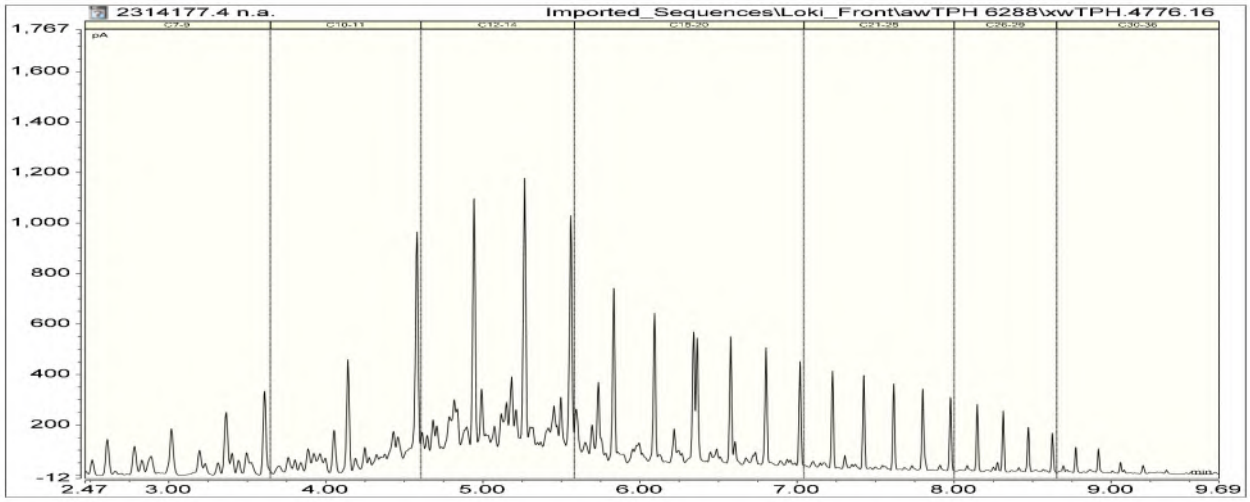
Client:	Greymouth Petroleum Limited	**Lab No:**	2314177 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	31-Jan-2020
	C/- Greymouth Petroleum Limited	**Date Reported:**	18-Feb-2020
	14 Connett Road West	**Quote No:**	81870
	Bell Block	**Order No:**	248844
	New Plymouth 4312	**Client Reference:**	Return Fluid Composite
		Submitted By:	Nakeysha Lammers

Sample Type: Saline

		Sample Name:	Turangi 8 - Stim 5 Composite Return Fluid				
		Lab Number:	2314177.4				
Individual Tests							
pH*	pH Units		8.0	-	-	-	-
Total Alkalinity*	g/m³ as $CaCO_3$		2,800	-	-	-	-
Analysis Temperature for Bicarbonate‡	°C		20	-	-	-	-
Bicarbonate‡	g/m³ at Analysis Temperature		2,760	-	-	-	-
Total Hardness*	g/m³ as $CaCO_3$		114	-	-	-	-
Electrical Conductivity (EC)*	mS/m		1,213	-	-	-	-
Total Dissolved Solids (TDS)*	g/m³		9,600	-	-	-	-
Dissolved Barium	g/m³		38	-	-	-	-
Dissolved Bromine	g/m³		8.8	-	-	-	-
Dissolved Calcium	g/m³		34	-	-	-	-
Dissolved Copper	g/m³		0.029	-	-	-	-
Dissolved Iron	g/m³		1.19	-	-	-	-
Dissolved Magnesium	g/m³		7	-	-	-	-
Dissolved Manganese	g/m³		4.0	-	-	-	-
Total Nickel	g/m³		0.050	-	-	-	-
Total Potassium	g/m³		175	-	-	-	-
Total Sodium	g/m³		3,000	-	-	-	-
Total Sulphur*	g/m³		29	-	-	-	-
Total Zinc	g/m³		0.065	-	-	-	-
Chloride*	g/m³		2,500	-	-	-	-
Nitrite-N	g/m³		< 0.10 #1	-	-	-	-
Nitrate-N	g/m³		< 0.10	-	-	-	-
Nitrate*	g/m³		< 0.5	-	-	-	-
Nitrate-N + Nitrite-N	g/m³		< 0.10 #1	-	-	-	-
Sulphate*	g/m³		88	-	-	-	-
Ethylene Glycol in Water*							
Ethylene glycol*	g/m³		8	-	-	-	-
Propylene Glycol in Water*							
Propylene glycol*	g/m³		< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*							
Methanol*	g/m³		< 2	-	-	-	-
BTEX in Water by Headspace GC-MS*							
Benzene*	g/m³		4.3	-	-	-	-
Toluene*	g/m³		9.2	-	-	-	-
Ethylbenzene*	g/m³		1.36	-	-	-	-
m&p-Xylene*	g/m³		9.2	-	-	-	-
o-Xylene*	g/m³		3.2	-	-	-	-

Sample Type: Saline						
Sample Name:	Turangi 8 - Stim 5 Composite Return Fluid					
Lab Number:	2314177.4					
Formaldehyde in Water by DNPH & LCMSMS*						
Formaldehyde* g/m³	< 0.15	-	-	-	-	
Total Petroleum Hydrocarbons in Water*						
C7 - C9* g/m³	183	-	-	-	-	
C10 - C14* g/m³	1,110	-	-	-	-	
C15 - C36* g/m³	1,000	-	-	-	-	
Total hydrocarbons (C7 - C36)* g/m³	2,300	-	-	-	-	

2314177.4
Turangi 8 - Stim 5 Composite Return Fluid
Client Chromatogram for TPH by FID



Analyst's Comments

‡ Analysis subcontracted to an external provider. Refer to the Summary of Methods section for more details.

#1 Due to the nature of this sample a dilution was performed prior to analysis, resulting in a detection limit higher than that normally achieved for the NO2N, NO3N and NOxN analysis.

Appendix No.1 - GNS report

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request.
Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	4
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	4
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m³	4
BTEX in Water by Headspace GC-MS*	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m³	4
Formaldehyde in Water by DNPH & LCMSMS*	DNPH derivatisation, extraction, LCMSMS	0.02 g/m³	4
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m³	4
Filtration, Unpreserved*	Sample filtration through 0.45µm membrane filter.	-	4
Total Digestion*	Boiling nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	4
Total Digestion of Saline Samples*	Nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	4

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
pH*	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	4
Total Alkalinity*	Saline water, Titration to pH 4.5.	1.0 g/m³ as $CaCO_3$	4
Analysis Temperature for Bicarbonate	Temperature at which Bicarbonate titration was conducted as reported by Geological & Nuclear Sciences, Wairakei.	1.0 °C	4
Bicarbonate	Bicarbonate (HCO3) Titration Method conducted at reported temperature. Subcontracted to Geological & Nuclear Sciences, Wairakei. ASTM Standards D513-82 Vol.11.01 of 1988.	20 g/m³ at Analysis Temperature	4
Total Hardness*	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m³ as $CaCO_3$	4
Electrical Conductivity (EC)*	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	4
Total Dissolved Solids (TDS)*	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	50 g/m³	4
Filtration for dissolved metals analysis*	Sample filtration through 0.45µm membrane filter and preservation with nitric acid. APHA 3030 B 23rd ed. 2017.	-	4
Dissolved Barium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0006 g/m³	4
Dissolved Bromine	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.10 g/m³	4
Dissolved Calcium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.0 g/m³	4
Dissolved Copper	Filtered sample, ICP-MS, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	4
Dissolved Iron	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.004 g/m³	4
Dissolved Magnesium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.4 g/m³	4
Dissolved Manganese	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m³	4
Total Nickel	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0070 g/m³	4
Total Potassium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.1 g/m³	4
Total Sodium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.42 g/m³	4
Total Sulphur*	Nitric acid digestion, ICP-OES (method may not fully account for H_2S due to volatilisation during digestion). All forms of oxidised and organic sulphur will be determined by this method. APHA 3120 B 23rd ed. 2017.	0.5 g/m³	4
Total Zinc	Nitric acid digestion, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0042 g/m³	4
Chloride*	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m³	4
Nitrite-N	Saline sample. Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m³	4
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m³	4
Nitrate*	Calculation from Nitrate-N.	0.005 g/m³	4
Nitrate-N + Nitrite-N	Saline sample. Total oxidised nitrogen. Automated cadmium reduction, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m³	4
Total Sulphate*	Calculation: from total sulphur.	2 g/m³	4

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

GNS SCIENCE TE PŪ AO
ANALYTICAL LABORATORY
Private Bag 2000, Taupo
Phone: (07) 374 8211
Fax: (07) 374 8199
Email: w.labmanager@gns.cri.nz

CERTIFICATE OF ANALYSIS

ENVSUBGNS_WAIRAKEI 114

Report No: 2020020407

Customer Ref:153311

Ara Heron
RJ Hill Laboratories (Hamilton)
Environmental Reports Officers
Private Bag 3205
Hamilton

GNS Lot No: 2020020407

GNS Sample No.	2020000669
Collection Date	
Site ID	2314177.4
Field ID	

pH		7.65	-	-	-
Bicarbonate (Total)	mg/l	2765	-	-	-
HCO_3 Analysis Temperature	°C	20	-	-	-
HCO_3 Analysis Date		04/02/2020	-	-	-

SUMMARY OF METHODS AND DETECTION LIMITS

The following table gives a brief description of the methods used to conduct the analyses on this report.
The detection limits given below are those attainable in a relatively clean matrix.

Parameter	Method	*Detection Limit	
Bicarbonate (total)	HCO_3 Titration Method ASTM Standards D513-82 Vol.11.01 1988	20	mg/l
pH	Electrometric Method - APHA 4500-H+ B 23rd Edition 2017	-	-

*Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis.

Notes: These samples were collected by yourselves (or your agent) and analysed as received at the laboratory. This report must not be reproduced, except in full, without the written consent of the signatory. Samples are held at the laboratory after reporting for a period of 2 to 6 months, dependent on sample type.

A Noddings
Ann Noddings
Senior Technician

ilac-MRA IANZ ACCREDITED LABORATORY Tests marked with a † are not accredited and are outside the scope of the laboratory's accreditation

Hill Laboratories
TRIED, TESTED AND TRUSTED

R J Hill Laboratories Limited
28 Duke Street Frankton 3204
Private Bag 3205
Hamilton 3240 New Zealand

T 0508 HILL LAB (44 555 22)
T +64 7 858 2000
E mail@hill-labs.co.nz
W www.hill-laboratories.com

Certificate of Analysis

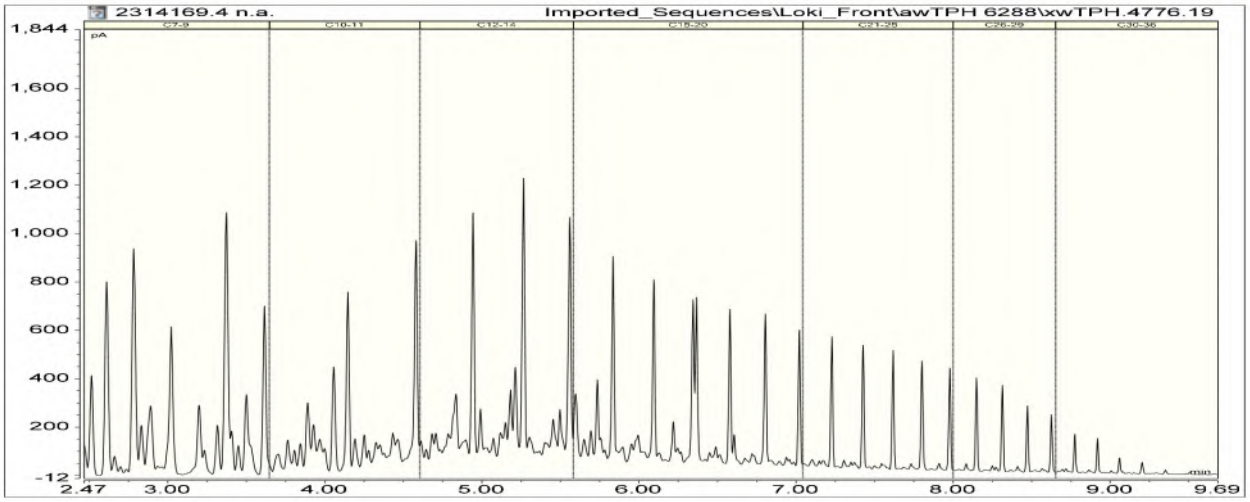
Client:	Greymouth Petroleum Limited	**Lab No:**	2314169 SPv1
Contact:	Nakeysha Lammers	**Date Received:**	31-Jan-2020
	C/- Greymouth Petroleum Limited	**Date Reported:**	18-Feb-2020
	14 Connett Road West	**Quote No:**	81870
	Bell Block	**Order No:**	249098
	New Plymouth 4312	**Client Reference:**	Return Fluid Composite
		Submitted By:	Nakeysha Lammers

Sample Type: Saline

		Sample Name:	Turangi 8 Stim 6 Composite Return Fluid				
		Lab Number:	2314169.4				
Individual Tests							
pH*	pH Units		7.1	-	-	-	-
Total Alkalinity*	g/m^3 as $CaCO_3$		2,300	-	-	-	-
Analysis Temperature for Bicarbonate‡	°C		22	-	-	-	-
Bicarbonate‡	g/m^3 at Analysis Temperature		1,979	-	-	-	-
Total Hardness*	g/m^3 as $CaCO_3$		102	-	-	-	-
Electrical Conductivity (EC)*	mS/m		1,071	-	-	-	-
Total Dissolved Solids (TDS)*	g/m^3		10,000	-	-	-	-
Dissolved Barium	g/m^3		16.7	-	-	-	-
Dissolved Bromine	g/m^3		9.3	-	-	-	-
Dissolved Calcium	g/m^3		30	-	-	-	-
Dissolved Copper	g/m^3		0.028	-	-	-	-
Dissolved Iron	g/m^3		8.5	-	-	-	-
Dissolved Magnesium	g/m^3		7	-	-	-	-
Dissolved Manganese	g/m^3		8.5	-	-	-	-
Total Nickel	g/m^3		1.27 #1	-	-	-	-
Total Potassium	g/m^3		169	-	-	-	-
Total Sodium	g/m^3		2,400	-	-	-	-
Total Sulphur*	g/m^3		25	-	-	-	-
Total Zinc	g/m^3		0.077	-	-	-	-
Chloride*	g/m^3		2,500	-	-	-	-
Nitrite-N	g/m^3		< 0.10 #2	-	-	-	-
Nitrate-N	g/m^3		< 0.10	-	-	-	-
Nitrate*	g/m^3		< 0.5	-	-	-	-
Nitrate-N + Nitrite-N	g/m^3		< 0.10 #2	-	-	-	-
Sulphate*	g/m^3		74	-	-	-	-
Ethylene Glycol in Water*							
Ethylene glycol*	g/m^3		5	-	-	-	-
Propylene Glycol in Water*							
Propylene glycol*	g/m^3		< 4	-	-	-	-
Methanol in Water - Aqueous Solvents*							
Methanol*	g/m^3		112	-	-	-	-
BTEX in Water by Headspace GC-MS*							
Benzene*	g/m^3		33	-	-	-	-
Toluene*	g/m^3		81	-	-	-	-
Ethylbenzene*	g/m^3		9.5	-	-	-	-
m&p-Xylene*	g/m^3		61	-	-	-	-
o-Xylene*	g/m^3		17.3	-	-	-	-

Sample Type: Saline						
Sample Name:	Turangi 8 Stim 6 Composite Return Fluid					
Lab Number:	2314169.4					
Formaldehyde in Water by DNPH & LCMSMS*						
Formaldehyde* g/m³	< 0.15	-	-	-	-	
Total Petroleum Hydrocarbons in Water*						
C7 - C9* g/m³	690	-	-	-	-	
C10 - C14* g/m³	1,170	-	-	-	-	
C15 - C36* g/m³	1,200	-	-	-	-	
Total hydrocarbons (C7 - C36)* g/m³	3,100	-	-	-	-	

2314169.4
Turangi 8 Stim 6 Composite Return Fluid
Client Chromatogram for TPH by FID



Analyst's Comments

‡ Analysis subcontracted to an external provider. Refer to the Summary of Methods section for more details.

#1 It should be noted that the replicate analyses performed on this sample as part of our in-house Quality Assurance procedures showed greater variation than would normally be expected. This may reflect the heterogeneity of the sample. The average of the results of the replicate analyses has been reported. Replicate 1 = 1.20mg/L, replicate 2 = 1.34mg/L.

#2 Due to the nature of this sample a dilution was performed prior to analysis, resulting in a detection limit higher than that normally achieved for the NO2N, NO3N and NOxN analysis.

Appendix No.1 - GNS report

Summary of Methods

The following table(s) gives a brief description of the methods used to conduct the analyses for this job. The detection limits given below are those attainable in a relatively simple matrix. Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis. A detection limit range indicates the lowest and highest detection limits in the associated suite of analytes. A full listing of compounds and detection limits are available from the laboratory upon request. Unless otherwise indicated, analyses were performed at Hill Laboratories, 28 Duke Street, Frankton, Hamilton 3204.

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Ethylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	4
Propylene Glycol in Water*	Direct injection, dual column GC-FID	4 g/m³	4
Methanol in Water - Aqueous Solvents*	Direct injection, dual column GC-FID	1.0 g/m³	4
BTEX in Water by Headspace GC-MS*	Headspace GC-MS analysis, US EPA 8260B [KBIs:26687,3629]	0.0010 - 0.002 g/m³	4
Formaldehyde in Water by DNPH & LCMSMS*	DNPH derivatisation, extraction, LCMSMS	0.02 g/m³	4
Total Petroleum Hydrocarbons in Water*	Hexane extraction, GC-FID analysis US EPA 8015B / MfE Petroleum Industry Guidelines	0.10 - 0.7 g/m³	4
Filtration, Unpreserved*	Sample filtration through 0.45µm membrane filter.	-	4
Total Digestion*	Boiling nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	4

Sample Type: Saline			
Test	**Method Description**	**Default Detection Limit**	**Sample No**
Total Digestion of Saline Samples*	Nitric acid digestion. APHA 3030 E (modified) 23rd ed. 2017.	-	4
pH*	pH meter. APHA 4500-H^+ B 23rd ed. 2017. Note: It is not possible to achieve the APHA Maximum Storage Recommendation for this test (15 min) when samples are analysed upon receipt at the laboratory, and not in the field. Samples and Standards are analysed at an equivalent laboratory temperature (typically 18 to 22 °C). Temperature compensation is used.	0.1 pH Units	4
Total Alkalinity*	Saline water, Titration to pH 4.5.	1.0 g/m^3 as $CaCO_3$	4
Analysis Temperature for Bicarbonate	Temperature at which Bicarbonate titration was conducted as reported by Geological & Nuclear Sciences, Wairakei.	1.0 °C	4
Bicarbonate	Bicarbonate (HCO3) Titration Method conducted at reported temperature. Subcontracted to Geological & Nuclear Sciences, Wairakei. ASTM Standards D513-82 Vol.11.01 of 1988.	20 g/m^3 at Analysis Temperature	4
Total Hardness*	Calculation from Calcium and Magnesium. APHA 2340 B 23rd ed. 2017.	1.0 g/m^3 as $CaCO_3$	4
Electrical Conductivity (EC)*	Conductivity meter, 25°C. APHA 2510 B 23rd ed. 2017.	0.1 mS/m	4
Total Dissolved Solids (TDS)*	Filtration through GF/C (1.2 µm), gravimetric. APHA 2540 C (modified; drying temperature of 103 - 105°C used rather than 180 ± 2°C) 23rd ed. 2017.	50 g/m^3	4
Filtration for dissolved metals analysis*	Sample filtration through 0.45µm membrane filter and preservation with nitric acid. APHA 3030 B 23rd ed. 2017.	-	4
Dissolved Barium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0006 g/m^3	4
Dissolved Bromine	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.10 g/m^3	4
Dissolved Calcium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.0 g/m^3	4
Dissolved Copper	Filtered sample, ICP-MS, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m^3	4
Dissolved Iron	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.004 g/m^3	4
Dissolved Magnesium	Filtered sample, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.4 g/m^3	4
Dissolved Manganese	Filtered sample, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0010 g/m^3	4
Total Nickel	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.0070 g/m^3	4
Total Potassium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	1.1 g/m^3	4
Total Sodium	Nitric acid digestion, ICP-MS, ultratrace level. APHA 3125 B 23rd ed. 2017.	0.42 g/m^3	4
Total Sulphur*	Nitric acid digestion, ICP-OES (method may not fully account for H_2S due to volatilisation during digestion). All forms of oxidised and organic sulphur will be determined by this method. APHA 3120 B 23rd ed. 2017.	0.5 g/m^3	4
Total Zinc	Nitric acid digestion, ICP-MS with dynamic reaction cell, ultratrace. APHA 3125 B 23rd ed. 2017.	0.0042 g/m^3	4
Chloride*	Filtered sample. Ion Chromatography. APHA 4110 B (modified) 23rd ed. 2017.	0.5 g/m^3	4
Nitrite-N	Saline sample. Automated Azo dye colorimetry, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m^3	4
Nitrate-N	Calculation: (Nitrate-N + Nitrite-N) - NO2N. In-House.	0.0010 g/m^3	4
Nitrate*	Calculation from Nitrate-N.	0.005 g/m^3	4
Nitrate-N + Nitrite-N	Saline sample. Total oxidised nitrogen. Automated cadmium reduction, Flow injection analyser. APHA 4500-NO_3^- I (modified) 23rd ed. 2017.	0.0010 g/m^3	4
Total Sulphate*	Calculation: from total sulphur.	2 g/m^3	4

These samples were collected by yourselves (or your agent) and analysed as received at the laboratory.

Dates of testing are available on request. Please contact the laboratory for more information.

Samples are held at the laboratory after reporting for a length of time based on the stability of the samples and analytes being tested (considering any preservation used), and the storage space available. Once the storage period is completed, the samples are discarded unless otherwise agreed with the customer. Extended storage times may incur additional charges.

Ara Heron BSc (Tech)
Client Services Manager - Environmental

GNS SCIENCE TE PŪ AO
ANALYTICAL LABORATORY
Private Bag 2000, Taupo
Phone: (07) 374 8211
Fax: (07) 374 8199
Email: w.labmanager@gns.cri.nz

CERTIFICATE OF ANALYSIS

ENVSUBGNS_WAIRAKEI 112

Report No: 2020020405

Customer Ref:153309

Ara Heron
RJ Hill Laboratories (Hamilton)
Environmental Reports Officers
Private Bag 3205
Hamilton

GNS Lot No: 2020020405

GNS Sample No.	2020000667
Collection Date	
Site ID	2314169.4
Field ID	

pH		7.44	-	-	-
Bicarbonate (Total)	mg/l	1979	-	-	-
HCO_3 Analysis Temperature	°C	22	-	-	-
HCO_3 Analysis Date		05/02/2020	-	-	-

SUMMARY OF METHODS AND DETECTION LIMITS

The following table gives a brief description of the methods used to conduct the analyses on this report.
The detection limits given below are those attainable in a relatively clean matrix.

Parameter	Method	*Detection Limit	
Bicarbonate (total)	HCO_3 Titration Method ASTM Standards D513-82 Vol.11.01 1988	20	mg/l
pH	Electrometric Method - APHA 4500-H+ B 23rd Edition 2017	-	-

*Detection limits may be higher for individual samples should insufficient sample be available, or if the matrix requires that dilutions be performed during analysis.

Notes: These samples were collected by yourselves (or your agent) and analysed as received at the laboratory. This report must not be reproduced, except in full, without the written consent of the signatory. Samples are held at the laboratory after reporting for a period of 2 to 6 months, dependent on sample type.

A Noddings
Ann Noddings
Senior Technician

ilac-MRA IANZ ACCREDITED LABORATORY Tests marked with a † are not accredited and are outside the scope of the laboratory's accreditation